

# Novel UV-C Emitting Phosphors for Hg Low Pressure Discharge Lamps

Jan-Niklas Keil, M.Sc., Mike Broxtermann, M.Sc., Prof. Dr. Thomas Jüstel  
Münster University of Applied Sciences, Department of Chemical Engineering

Stegerwaldstrasse 39, D-48565 Steinfurt, Germany

Corresponding authors: keil@fh-muenster.de and tj@fh-muenster.de

20<sup>th</sup> JCF-Frühjahrssymposium, March 2018, Konstanz, Germany

## Introduction

Water disinfection with UV-C radiation gets more and more important over the last 20 years. Hg-low-pressure discharge lamps are typically used as artificial UV-C sources. The main emission (85%) of these lamps lays at 254 nm, 12% lays at 185 nm (VUV-region) and 3% in the visible range of the electromagnetic spectrum. Therefore, the majority of the emitted radiation is UV-C light, which can be used directly for water disinfection. The 12% of VUV emission, become absorbed by water under formation of OH-radicals, already in small layer thicknesses. To increase the disinfection efficiency of Hg-low-pressure discharge lamps it is necessary to convert the emitted VUV-radiation into UV-C-radiation. Therefore phosphors are needed, which can be excited with 185 nm and emits radiation between 200 and 310 nm. In addition to disinfection purposes, UV-C radiation can also be used for so called "advanced oxidation processes" (AOP). AOP is a fast growing field in the area of water treatment, in which OH-radicals are used to break C-C and C-H bonds to decompose organic pollutants [1].

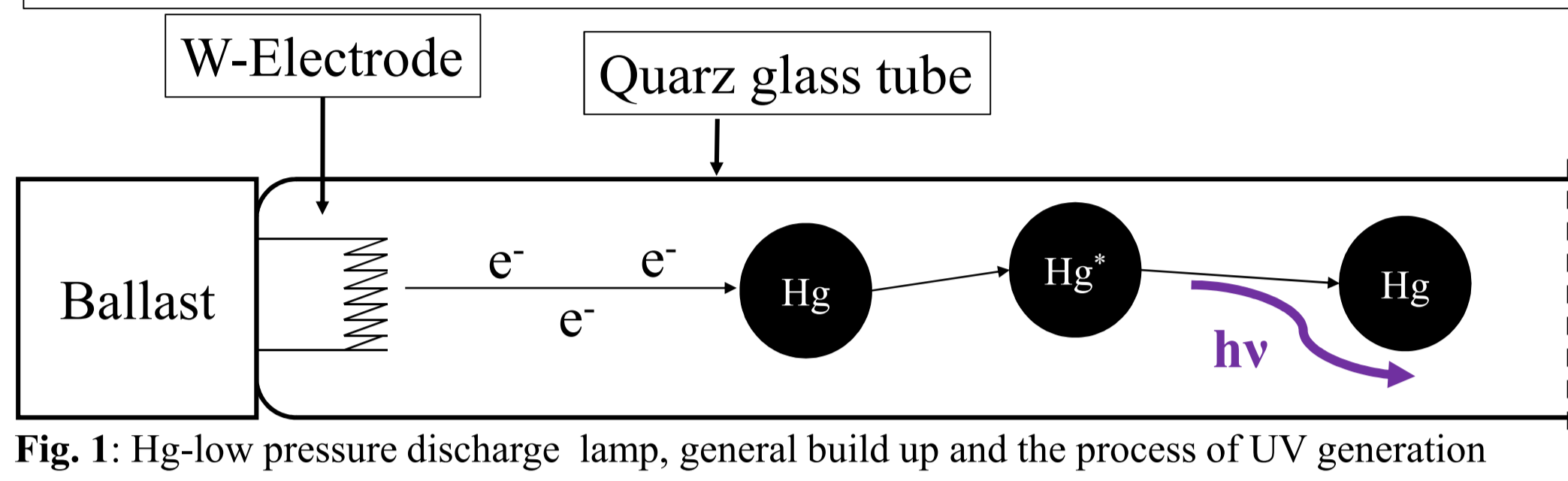


Fig. 1: Hg-low pressure discharge lamp, general build up and the process of UV generation

## Processes inside the gas discharge

1. Emission of electrons: Cathode  $\rightarrow e^-$
2. Excitation of Hg-atoms:  $Hg + e^- \rightarrow Hg^* + e^-$
3. Relaxation of excited Hg-atoms:  $Hg^* \rightarrow Hg + hv$

## Prerequisites for potential UV-C emitting phosphors

- Suitable band gap ( $> 6$  eV)
- High thermal, photo and chemical stability
- Redox stable activators
- High absorption cross section around 185 nm
- Emission band between 200 and 280 nm

## Examples for suitable host materials and activators

- |                 |             |
|-----------------|-------------|
| Host materials: | Activators: |
| - Phosphates    | - $Pr^{3+}$ |
| - Borates       | - $Nd^{3+}$ |
| - Aluminates    | - $Bi^{3+}$ |
| - Silicates     |             |

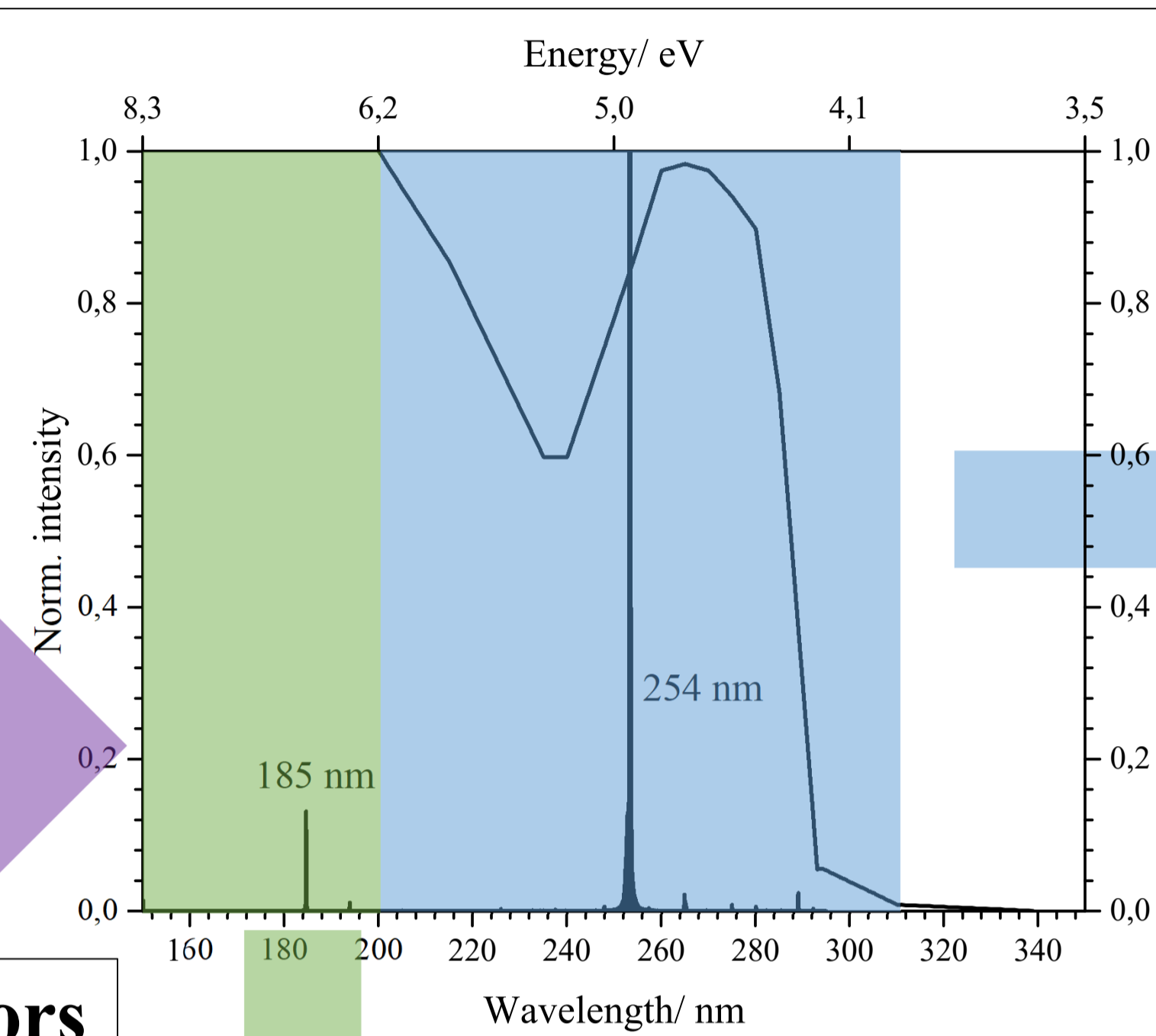
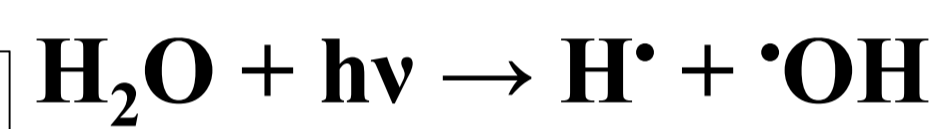
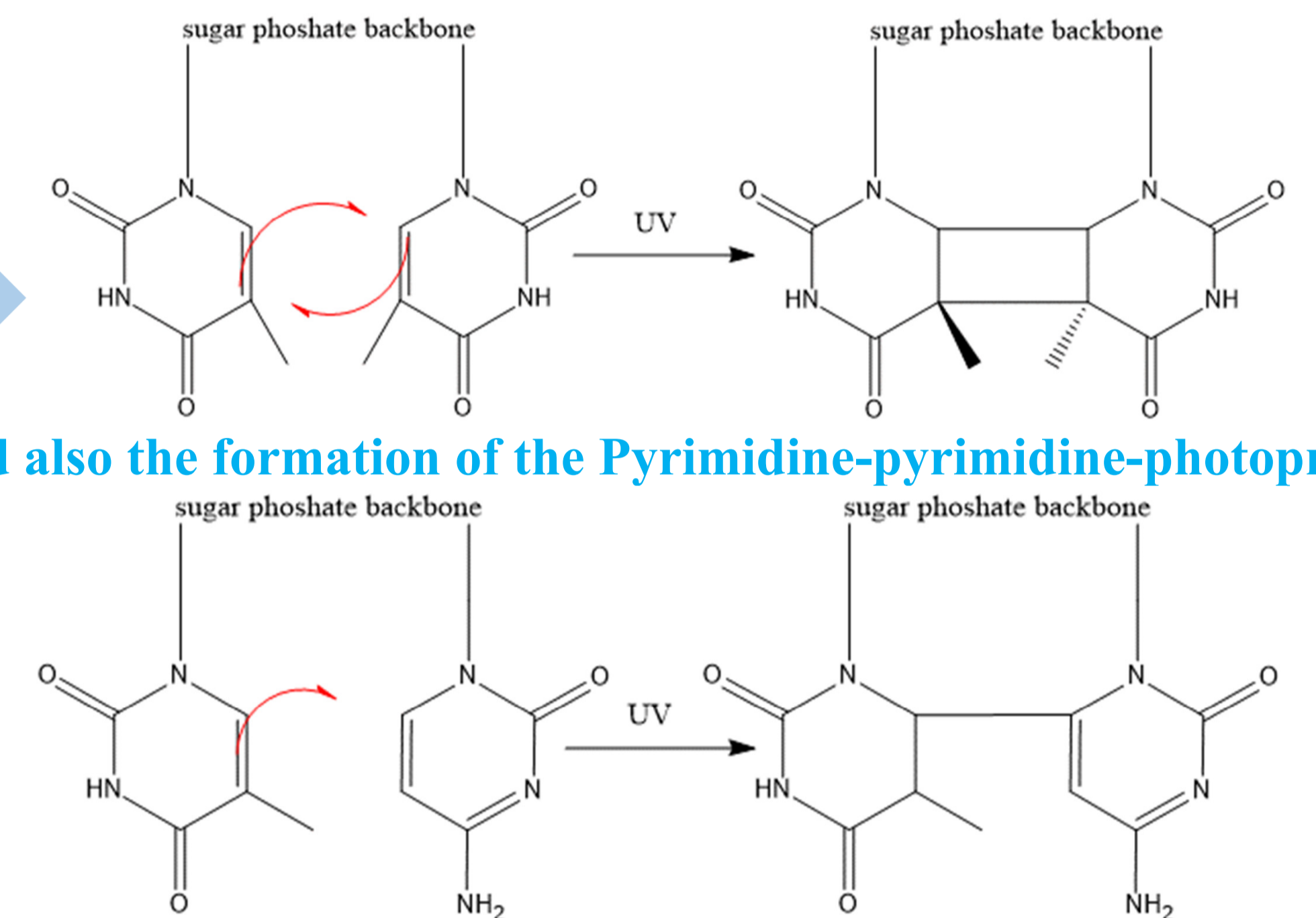


Fig. 2: Emission spectrum of a Hg-low pressure discharge lamp and germicidal action curve for *B. subtilis* (DIN 5031-10)

Radiation under 200 nm induces the photolysis of water



Radiation between 200 and 310 nm induces the photochemical dimerization of two adjacent thymine bases by [2+2]-cycloaddition



and also the formation of the Pyrimidine-pyrimidine-photoproduct

## „Advanced oxidation processes“

1. Formation of hydroxyl radicals out of hydrogen peroxide:  $H_2O_2 + hv \rightarrow 2 \cdot OH$
2. Cleavage of C-C and C-H bonds via hydrogen abstraction or addition reactions [2]:
  - a.  $\cdot OH + H_3C-OH \rightarrow H_2O + H_2C^{\cdot}-OH$
  - b.  $H_2C^{\cdot}-OH + \cdot OH + O_2 \rightarrow H_2C=O + H_2O$
  - c.  $H_2C=O + \cdot OH + O_2 \rightarrow HCOOH$
  - d.  $HCOOH + \cdot OH + O_2 \rightarrow CO_2 + H_2O$
 or
  - a.  $H_2C=CH_2 + \cdot OH \rightarrow H_2C^{\cdot}-CH_2-OH$
  - b.  $H_2C^{\cdot}-CH_2-OH + \cdot OH \rightarrow HO-H_2C-CH_2-OH$

## Examples for the photoluminescence of UV-C emitting phosphors

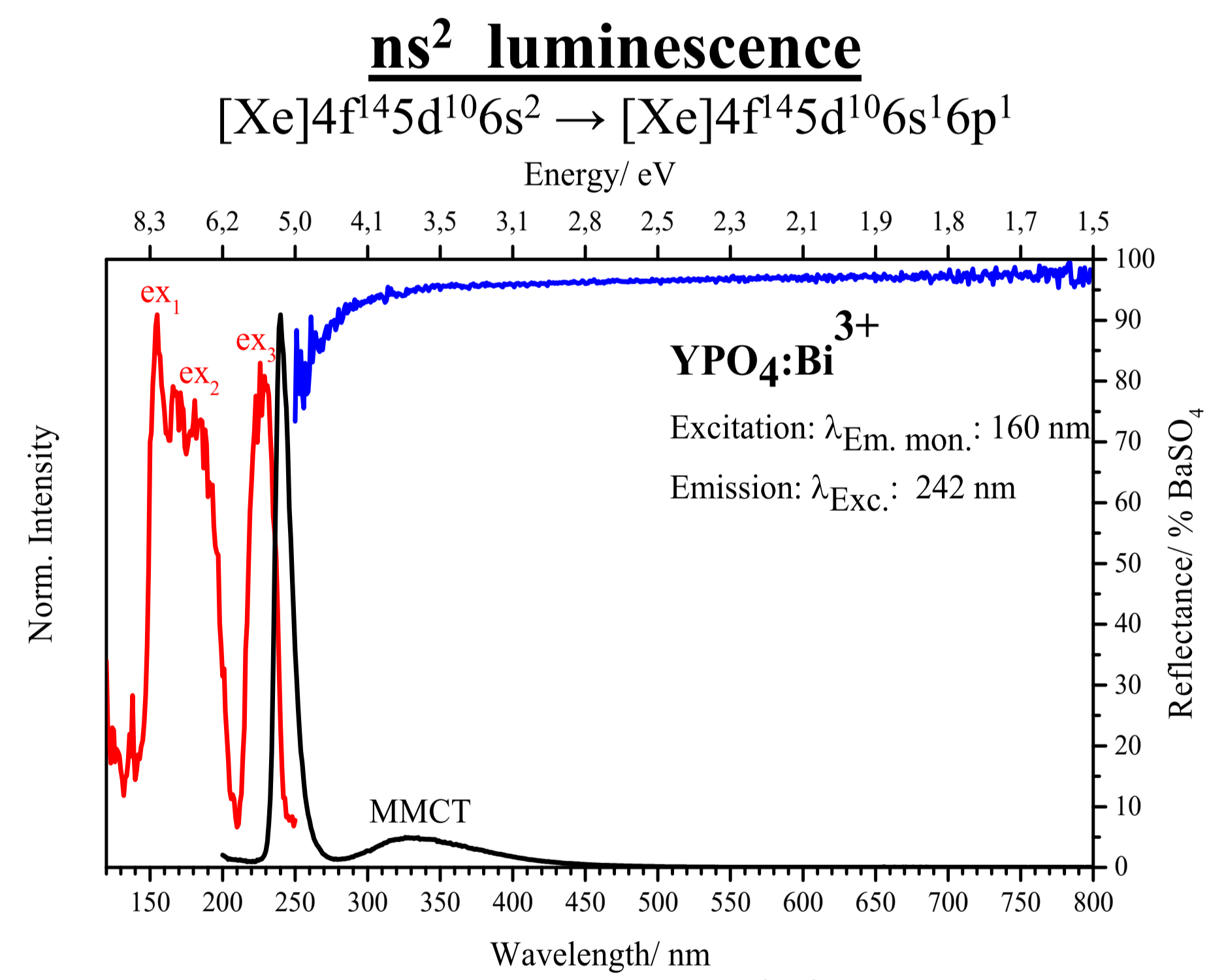


Fig. 3: Room temperature PLE, PL and reflectance spectra of  $YPO_4:Bi^{3+}$  and  $YPO_4:Pr^{3+}$

## 4f-5d luminescence

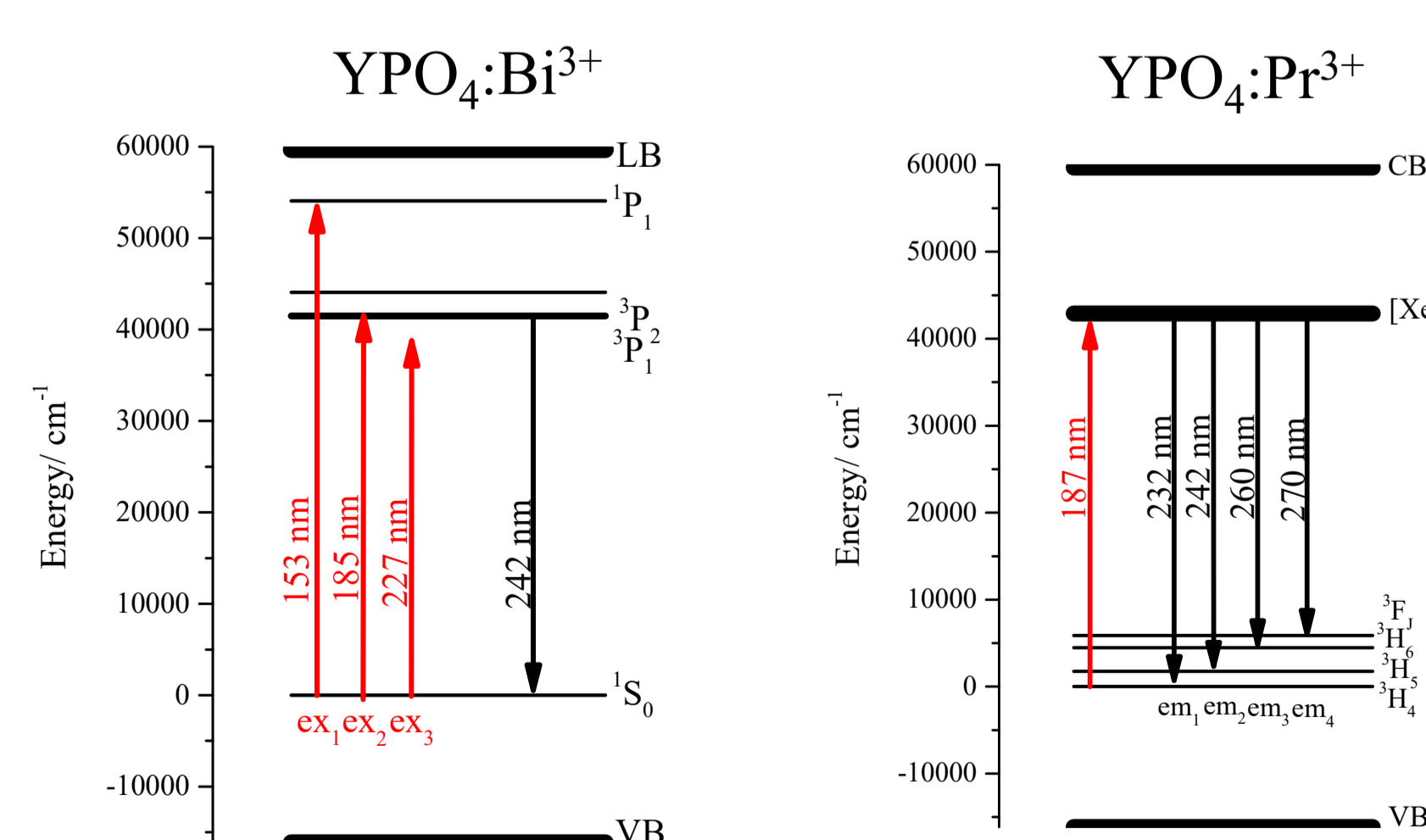
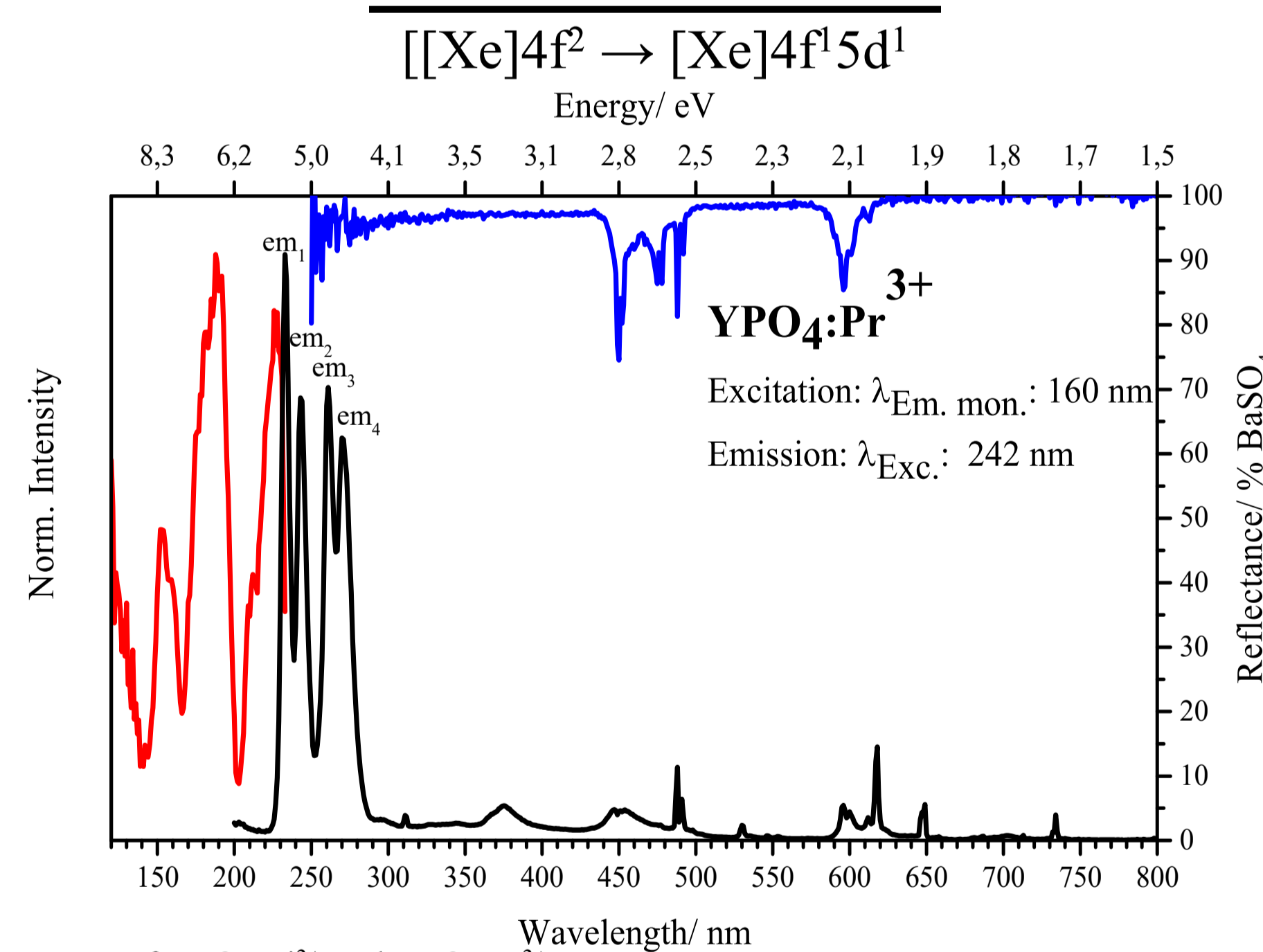


Fig. 4: Energy level diagram of  $YPO_4:Bi^{3+}$  and  $YPO_4:Pr^{3+}$  [3,4]

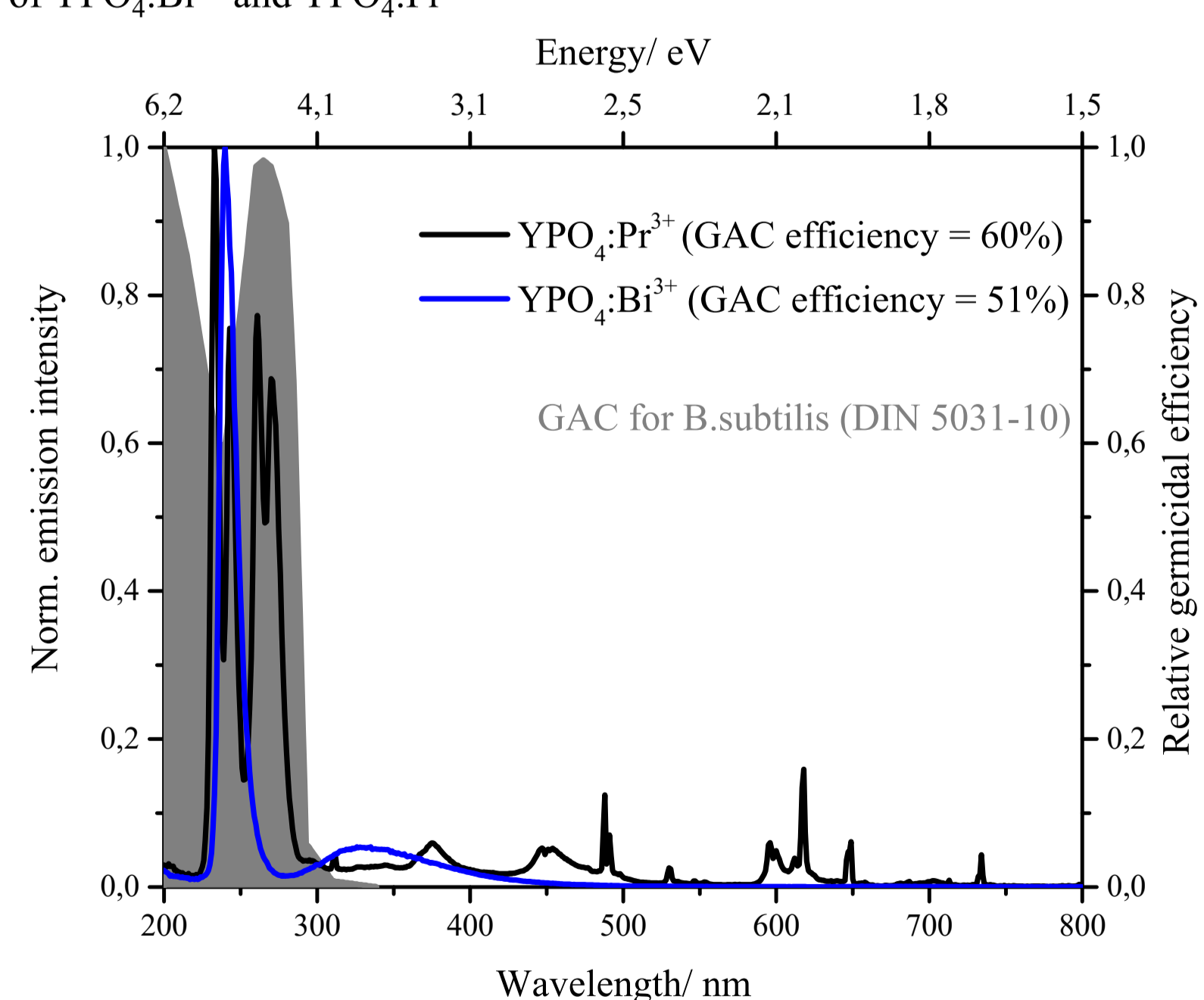


Fig. 5 Relative germicidal efficacy for *B. subtilis* according DIN 5031-10 and photoluminescence emission spectra of  $YPO_4:Bi^{3+}$ ,  $YPO_4:Pr^{3+}$  ( $\lambda_{Exc.} = 160$  nm)

## Example for the efficiency increase

Common values for big disinfection devices:

- Lamp power: 50 kW
- Efficiency 185 nm: 10 %
- Efficiency 254 nm: 40 %

Without phosphor conversion of the 185 nm emission:

**20 kW (40%) for disinfection useable**

With phosphor conversion of the 185 nm emission, using a hypothetical phosphor with a quantum efficiency of 90%:

**24.5 kW (49%) for disinfection useable**

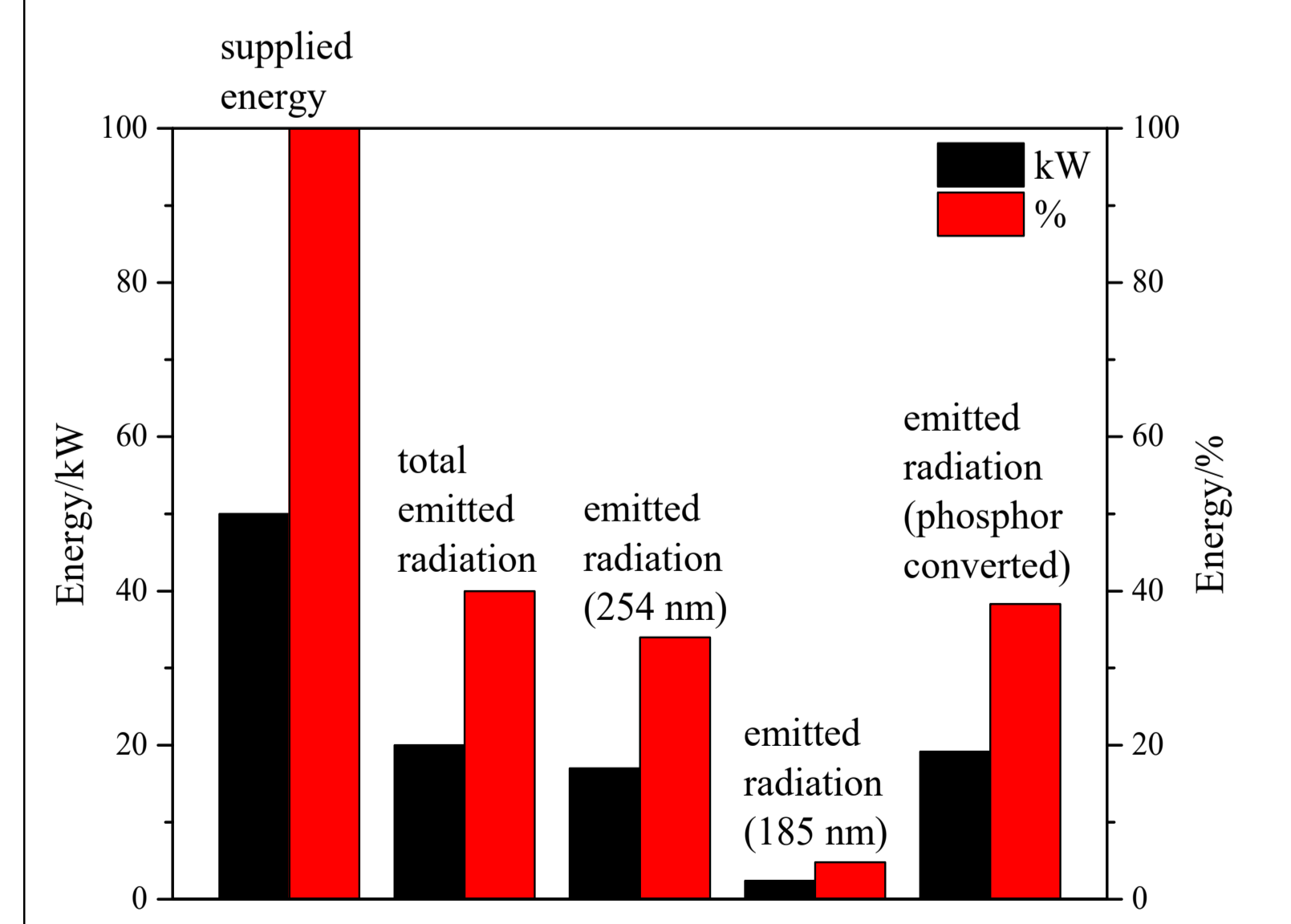


Fig. 6: Overview energy distribution for a big disinfection device with and without phosphor conversion of the 185 nm line

## Conclusions

Future work will deal with the more UV-C emitting phosphors. The herein presented phosphors are already known and multiple studies were dedicated to them.

## Literature

- [1] R. Andreozzi et al., *Cat. Today*, 53, 2008, 51-59
- [2] System SE. *The UV/oxidation handbook*, 1994
- [3] R.H.P. Awater, P. Dorenbos, *J. Lumi.*, 184, 2017, 221-231
- [4] A.J.J. Bos et al., *Rad. Measure.*, 43, 2008, 222-226

