

# LFDFT Calculations of Praseodymium doped binary Fluorides compared with Experimental Results



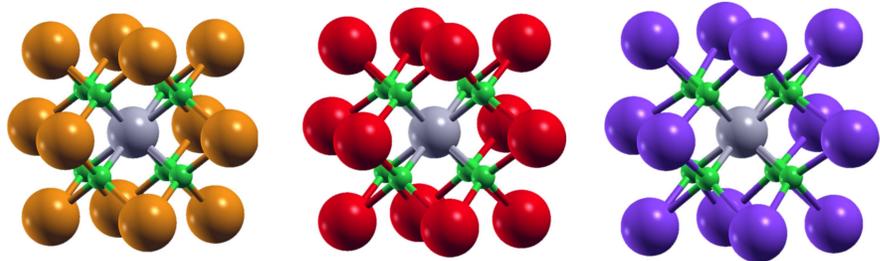
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## Background

Research efforts for new materials as luminescent pigments for lighting or solid state lasers are currently growing world-wide. An activator ion which exhibits the potential for all these applications is trivalent praseodymium ( $\text{Pr}^{3+}$ ). The emission spectrum of  $\text{Pr}^{3+}$  depends strongly on the host structure and can show  $[\text{Xe}]4f^{15}d^1 \rightarrow [\text{Xe}]4f^2$  band emission solely in the UV range, or  $[\text{Xe}]4f^2 \rightarrow [\text{Xe}]4f^2$  line emission ranging from the UV to the NIR.



In this study, we consider the system  $\text{M}^{\text{II}}\text{F}_2:\text{Pr}^{3+}$  ( $\text{M}^{\text{II}} = \text{Ca}, \text{Sr}, \text{Ba}$ ) which exhibit the cubic  $\text{CaF}_2$  structure type (Fm-3m) and having  $\text{O}_h$  symmetry for the cation positions.

The energy multiplet levels were calculated using a model based on Ligand Field Theory, respectively Angular Overlap Model (AOM), and Density Functional Theory (DFT), named LFDFT<sup>1</sup>.

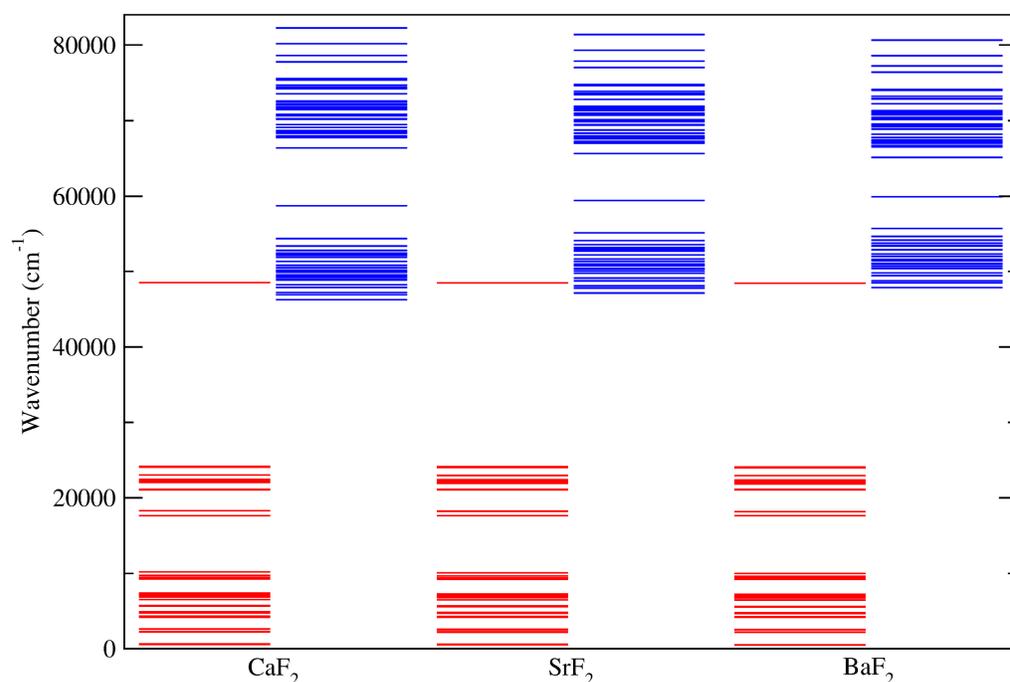
Sample preparation was performed by a precipitation route followed by a calcination step. Single phase luminescent materials were characterised by luminescence spectroscopy from the UV to the NIR part of the spectral range.

**Fig. 1** Crystal structures represent the central  $\text{Pr}^{3+}$  ion (grey) surrounded by eight  $\text{F}^-$  (green) and the second coordination sphere containing  $\text{Ca}^{2+}$ ,  $\text{Sr}^{2+}$  and  $\text{Ba}^{2+}$  in orange red and purple, respectively

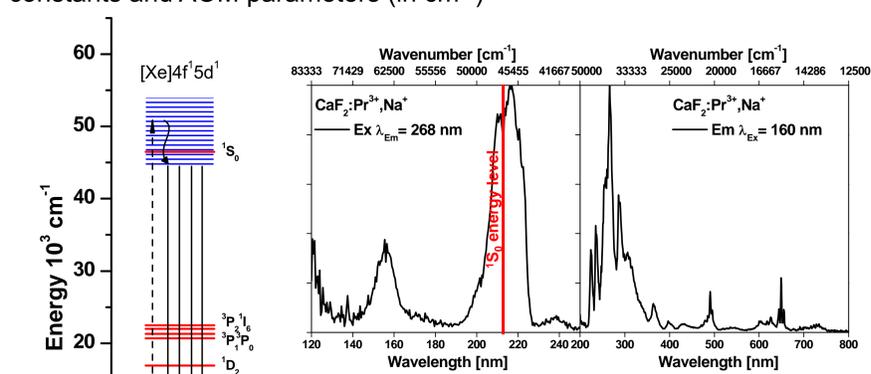
host	$\text{CaF}_2$	$\text{SrF}_2$	$\text{BaF}_2$
symmetry	$\text{O}_h$	$\text{O}_h$	$\text{O}_h$
$F_2(\text{ff})$	322.8	322.9	322.9
$F_4(\text{ff})$	41.7	41.7	41.7
$F_6(\text{ff})$	4.4	4.4	4.4
$G_1(\text{fd})$	358.9	356.9	351.2
$F_2(\text{fd})$	226.7	225.3	222.7
$G_3(\text{fd})$	30.7	30.5	30.1
$F_4(\text{fd})$	17.4	17.3	17.1
$G_5(\text{fd})$	4.8	4.7	4.7
$\zeta_{4f}$	710	710	710
$\zeta_{5d}$	945	945	945
$e_\sigma(f)$	599	538	491
$e_\pi(f)$	257	232	213
$e_\sigma(d)$	14114	13009	12154
$e_\pi(d)$	4705	4336	4051
$\Delta_{\text{AOM}}(\text{fd})$	15183	17918	20030

**Table 1** Electrostatic parameters, spin-orbit coupling constants and AOM parameters (in  $\text{cm}^{-1}$ )<sup>2</sup>

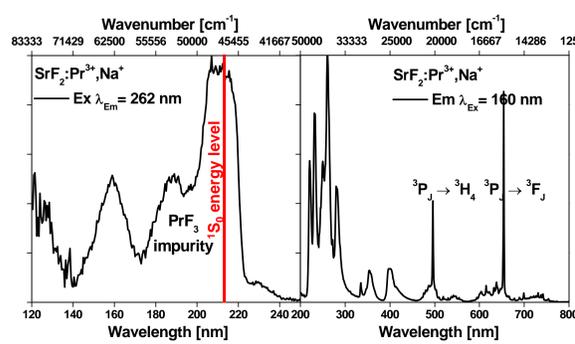
- Structure obtained from a periodical calculation using Density Functional Theory (DFT) with the VASP code
- Electrostatic and Ligand Field parameters obtained from a cluster calculation of  $\text{Pr}^{3+}$  and its surroundings with the Amsterdam Density Functional (ADF 2013) code
- The spin-orbit coupling constants for the 4f and 5d orbitals obtained using the relativistic ZORA approach
- LFDFT method to obtain the  $[\text{Xe}]4f^2$  and  $[\text{Xe}]4f^{15}d^1$  multiplet energies



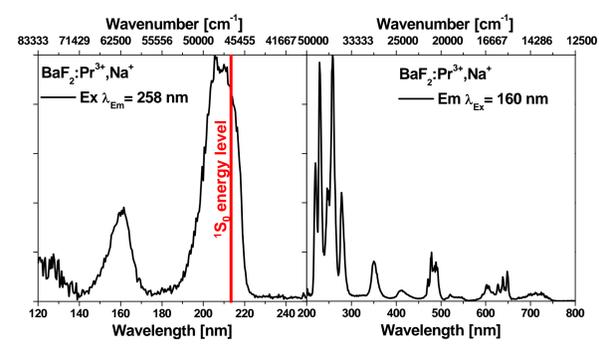
**Fig. 2** Calculated multiplet energy levels of the ground  $[\text{Xe}]4f^2$  (red) and the excited  $[\text{Xe}]4f^{15}d^1$  configurations (blue) for  $\text{Pr}^{3+}$  doped  $\text{CaF}_2$ ,  $\text{SrF}_2$ , and  $\text{BaF}_2$



**Fig. 4** Luminescence spectra of  $\text{CaF}_2:\text{Pr}^{3+}, \text{Na}^+$



**Fig. 5** Luminescence spectra of  $\text{SrF}_2:\text{Pr}^{3+}, \text{Na}^+$



**Fig. 6** Luminescence spectra of  $\text{BaF}_2:\text{Pr}^{3+}, \text{Na}^+$

The excitation spectra show two separated bands which can be compared to the  $e_g$  and  $t_{2g}$  levels of the excited  $[\text{Xe}]4f^{15}d^1$  configuration known for  $\text{O}_h$  symmetry. The highest energy level  $^1S_0$  of the ground  $[\text{Xe}]4f^2$  configuration is located at  $47000 \text{ cm}^{-1}$ . Therefore, the lowest energy level of the  $[\text{Xe}]4f^{15}d^1$  configuration is in all cases overlapping with the highest energy level  $^1S_0$ . The emission spectra are dominated by broad band emission from the lowest energy level of the  $[\text{Xe}]4f^{15}d^1$  configuration in the UV part of the spectral range.

## Conclusions

We predict and measure the optical behaviour of  $\text{Pr}^{3+}$  in the alkaline earth fluorides,  $\text{CaF}_2$ ,  $\text{SrF}_2$  and  $\text{BaF}_2$ . We use a validated model based on Ligand Field and Density Functional Theory (LFDFT) to calculate the multiplet energy levels arising from the ground  $[\text{Xe}]4f^2$  and excited  $[\text{Xe}]4f^{15}d^1$  configurations of  $\text{Pr}^{3+}$  in their chemical environment. Moreover, we characterize the considered materials by luminescence spectroscopy. In overall the theoretical determination corroborates to the experimental findings.

## Literature

<sup>1</sup> Ramanantoanina, H., Urland, W., García-Fuente, A., Cimpoesu, F. and Daul, C. Ligand field density functional theory for the prediction of future domestic lighting. Phys. Chem. Chem. Phys. 16, 14625–14634 (2014)

<sup>2</sup> Herden, B., García-Fuente, A., Ramanantoanina, H., Jüstel, T., Urland, W., and Daul, C. Photon Cascade Emission in  $\text{Pr}^{3+}$  doped fluorides with  $\text{CaF}_2$ -Structure: Application of a Model for its prediction. to be submitted