

## Electrochemische Reihe

#	Redoxreaktion	E0 [V] vs NHE
1	$\text{Sr}^+ + \text{e}^- = \text{Sr}$	-4,1
2	$\text{Ca}^+ + \text{e}^- = \text{Ca}$	-3,8
3	$\text{Pr}^{3+} + \text{e}^- = \text{Pr}^{2+}$	-3,1
4	$3 \text{N}_2 + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{HN}_3$	-3,09
5	$\text{Li}^+ + \text{e}^- = \text{Li}$	-3,0401
6	$\text{Cs}^+ + \text{e}^- = \text{Cs}$	-3,026
7	$\text{Ca}(\text{OH})_2 + 2 \text{e}^- = \text{Ca} + 2 \text{OH}^-$	-3,02
8	$\text{Er}^{3+} + \text{e}^- = \text{Er}^{2+}$	-3
9	$\text{Ba}(\text{OH})_2 + 2 \text{e}^- = \text{Ba} + 2 \text{OH}^-$	-2,99
10	$\text{Rb}^+ + \text{e}^- = \text{Rb}$	-2,98
11	$\text{K}^+ + \text{e}^- = \text{K}$	-2,931
12	$\text{Ba}^{2+} + 2 \text{e}^- = \text{Ba}$	-2,912
13	$\text{La}(\text{OH})_3 + 3 \text{e}^- = \text{La} + 3 \text{OH}^-$	-2,9
14	$\text{Fr}^+ + \text{e}^- = \text{Fr}$	-2,9
15	$\text{Sr}^{2+} + 2 \text{e}^- = \text{Sr}$	-2,899
16	$\text{Sr}(\text{OH})_2 + 2 \text{e}^- = \text{Sr} + 2 \text{OH}^-$	-2,88
17	$\text{Ca}^{2+} + 2 \text{e}^- = \text{Ca}$	-2,868
18	$\text{Eu}^{2+} + 2 \text{e}^- = \text{Eu}$	-2,812
19	$\text{Ra}^{2+} + 2 \text{e}^- = \text{Ra}$	-2,8
20	$\text{Ho}^{3+} + \text{e}^- = \text{Ho}^{2+}$	-2,8
21	$\text{Bk}^{3+} + \text{e}^- = \text{Bk}^{2+}$	-2,8
22	$\text{Yb}^{2+} + 2 \text{e}^- = \text{Yb}$	-2,76
23	$\text{Na}^+ + \text{e}^- = \text{Na}$	-2,71
24	$\text{Nd}^{3+} + \text{e}^- = \text{Nd}^{2+}$	-2,7
25	$\text{Mg}^+ + \text{e}^- = \text{Mg}$	-2,7
26	$\text{Mg}(\text{OH})_2 + 2 \text{e}^- = \text{Mg} + 2 \text{OH}^-$	-2,69
27	$\text{Sm}^{2+} + 2 \text{e}^- = \text{Sm}$	-2,68
28	$\text{Be}_2\text{O}_3^{2-} + 3 \text{H}_2\text{O} + 4 \text{e}^- = 2 \text{Be} + 6 \text{OH}^-$	-2,63
29	$\text{Pm}^{3+} + \text{e}^- = \text{Pm}^{2+}$	-2,6
30	$\text{Dy}^{3+} + \text{e}^- = \text{Dy}^{2+}$	-2,6
31	$\text{No}^{2+} + 2 \text{e}^- = \text{No}$	-2,5
32	$\text{HfO}(\text{OH})_2 + \text{H}_2\text{O} + 4 \text{e}^- = \text{Hf} + 4 \text{OH}^-$	-2,5
33	$\text{Th}(\text{OH})_4 + 4 \text{e}^- = \text{Th} + 4 \text{OH}^-$	-2,48
34	$\text{Tm}^{2+} + 2 \text{e}^- = \text{Tm}$	-2,4
35	$\text{Md}^{2+} + 2 \text{e}^- = \text{Md}$	-2,4
36	$\text{La}^{3+} + 3 \text{e}^- = \text{La}$	-2,379
37	$\text{Y}^{3+} + 3 \text{e}^- = \text{Y}$	-2,372
38	$\text{Mg}^{2+} + 2 \text{e}^- = \text{Mg}$	-2,372
39	$\text{ZrO}(\text{OH})_2 + \text{H}_2\text{O} + 4 \text{e}^- = \text{Zr} + 4 \text{OH}^-$	-2,36
40	$\text{Pr}^{3+} + 3 \text{e}^- = \text{Pr}$	-2,353
41	$\text{Ce}^{3+} + 3 \text{e}^- = \text{Ce}$	-2,336
42	$\text{Er}^{3+} + 3 \text{e}^- = \text{Er}$	-2,331
43	$\text{Ho}^{3+} + 3 \text{e}^- = \text{Ho}$	-2,33
44	$\text{H}_2\text{AlO}_3^- + \text{H}_2\text{O} + 3 \text{e}^- = \text{Al} + 4 \text{OH}^-$	-2,33
45	$\text{Nd}^{3+} + 3 \text{e}^- = \text{Nd}$	-2,323
46	$\text{Tm}^{3+} + 3 \text{e}^- = \text{Tm}$	-2,319
47	$\text{Al}(\text{OH})_3 + 3 \text{e}^- = \text{Al} + 3 \text{OH}^-$	-2,31
48	$\text{Sm}^{3+} + 3 \text{e}^- = \text{Sm}$	-2,304
49	$\text{Pm}^{3+} + 3 \text{e}^- = \text{Pm}$	-2,3
50	$\text{Fm}^{2+} + 2 \text{e}^- = \text{Fm}$	-2,3

51	$AM^{3+} + e^- = AM^{2+}$	-2,3
52	$Dy^{3+} + 3 e^- = Dy$	-2,295
53	$Tb^{3+} + 3 e^- = Tb$	-2,28
54	$Lu^{3+} + 3 e^- = Lu$	-2,28
55	$Gd^{3+} + 3 e^- = Gd$	-2,279
56	$H_2 + 2 e^- = 2 H^-$	-2,23
57	$Tm^{3+} + e^- = Tm^{2+}$	-2,2
58	$Pm^{2+} + 2 e^- = Pm$	-2,2
59	$Dy^{2+} + 2 e^- = Dy$	-2,2
60	$Ac^{3+} + 3 e^- = Ac$	-2,2
61	$Yb^{3+} + 3 e^- = Yb$	-2,19
62	$Es^{2+} + 2 e^- = Es$	-2,13
63	$Cf^{2+} + 2 e^- = Cf$	-2,12
64	$Nd^{2+} + 2 e^- = Nd$	-2,1
65	$Ho^{2+} + 2 e^- = Ho$	-2,1
66	$Sc^{3+} + 3 e^- = Sc$	-2,077
67	$AlF_6^{3-} + 3 e^- = Al + 6F^-$	-2,069
68	$AM^{3+} + 3 e^- = AM$	-2,048
69	$Cm^{3+} + 3 e^- = Cm$	-2,04
70	$Pu^{3+} + 3 e^- = Pu$	-2,031
71	$Pr^{2+} + 2 e^- = Pr$	-2
72	$Er^{2+} + 2 e^- = Er$	-2
73	$Eu^{3+} + 3 e^- = Eu$	-1,991
74	$Lr^{3+} + 3 e^- = Lr$	-1,96
75	$Cf^{3+} + 3 e^- = Cf$	-1,94
76	$Es^{3+} + 3 e^- = Es$	-1,91
77	$Pa^{4+} + e^- = Pa^{3+}$	-1,9
78	$AM^{2+} + 2 e^- = AM$	-1,9
79	$Th^{4+} + 4 e^- = Th$	-1,899
80	$Fm^{3+} + 3 e^- = Fm$	-1,89
81	$Np^{3+} + 3 e^- = Np$	-1,856
82	$Be^{2+} + 2 e^- = Be$	-1,847
83	$H_2P_2^- + e^- = P + 2 OH^-$	-1,82
84	$U^{3+} + 3 e^- = U$	-1,798
85	$Sr^{2+} + 2 e^- = Sr(Hg)$	-1,793
86	$H_2BO_3^- + H_2O + 3 e^- = B + 4 OH^-$	-1,79
87	$ThO_2 + 4 H^+ + 4 e^- = Th + 2 H_2O$	-1,789
88	$HfO_2 + 2 H^+ + 4 e^- = Hf + H_2O$	-1,724
89	$HPO_3^{2-} + 2 H_2O + 3 e^- = P + 5 OH^-$	-1,71
90	$SiO_3^{2-} + 3 H_2O + 4 e^- = Si + 6 OH^-$	-1,697
91	$Al^{3+} + 3 e^- = Al$	-1,662
92	$HPO_3^{2-} + 2 H_2O + 2 e^- = H_2PO_2^- + 3 OH^-$	-1,65
93	$Md^{3+} + 3 e^- = Md$	-1,65
94	$Ti^{2+} + 2 e^- = Ti$	-1,63
95	$Cf^{3+} + e^- = Cf^{2+}$	-1,6
96	$Bk^{2+} + 2 e^- = Bk$	-1,6
97	$Ba^{2+} + 2 e^- = Ba(Hg)$	-1,57
98	$Mn(OH)_2 + 2 e^- = Mn + 2 OH^-$	-1,56
99	$ZrO_2 + 4 H^+ + 4 e^- = Zr + 2 H_2O$	-1,553
100	$Sm^{3+} + e^- = Sm^{2+}$	-1,55
101	$Hf^{4+} + 4 e^- = Hf$	-1,55
102	$HfO_2 + 4 H^+ + 4 e^- = Hf + 2 H_2O$	-1,505

103	$\text{Pa}^{4+} + 4 \text{e}^- = \text{Pa}$	-1,49
104	$\text{Cr}(\text{OH})_3 + 3 \text{e}^- = \text{Cr} + 3 \text{OH}^-$	-1,48
105	$\text{Zr}^{4+} + 4 \text{e}^- = \text{Zr}$	-1,45
106	$\text{UO}_2^{2+} + 4 \text{H}^+ + 6 \text{e}^- = \text{U} + 2 \text{H}_2\text{O}$	-1,444
107	$\text{Ce}^{3+} + 3 \text{e}^- = \text{Ce}(\text{Hg})$	-1,4373
108	$\text{Ti}^{3+} + 3 \text{e}^- = \text{Ti}$	-1,37
109	$\text{Pa}^{3+} + 3 \text{e}^- = \text{Pa}$	-1,34
110	$\text{Es}^{3+} + \text{e}^- = \text{Es}^{2+}$	-1,3
111	$\text{ZnO} + \text{H}_2\text{O} + 2 \text{e}^- = \text{Zn} + 2 \text{OH}^-$	-1,26
112	$\text{Zn}(\text{OH})_2 + 2 \text{e}^- = \text{Zn} + 2 \text{OH}^-$	-1,249
113	$\text{SiF}_6^{2-} + 4 \text{e}^- = \text{Si} + 6 \text{F}^-$	-1,24
114	$\text{H}_2\text{BO}_3^- + 5 \text{H}_2\text{O} + 8 \text{e}^- = \text{BH}_4^- + 8 \text{OH}^-$	-1,24
115	$\text{H}_2\text{GaO}_3 + \text{H}_2\text{O} + 3 \text{e}^- = \text{Ga} + 4 \text{OH}^-$	-1,219
116	$\text{ZnO}_2^{2-} + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{Zn} + 4 \text{OH}^-$	-1,215
117	$\text{No}^{3+} + 3 \text{e}^- = \text{No}$	-1,2
118	$\text{CrO}_2^- + 2 \text{H}_2\text{O} + 3 \text{e}^- = \text{Cr} + 4 \text{OH}^-$	-1,2
119	$\text{Zn}(\text{OH})_4^{2-} + 2 \text{e}^- = \text{Zn} + 4 \text{OH}^-$	-1,199
120	$\text{Mn}^{2+} + 2 \text{e}^- = \text{Mn}$	-1,185
121	$\text{V}^{2+} + 2 \text{e}^- = \text{V}$	-1,175
122	$\text{Te} + 2 \text{e}^- = \text{Te}^{2-}$	-1,143
123	$2 \text{SO}_3^{2-} + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{S}_2\text{O}_4^{2-} + 4 \text{OH}^-$	-1,12
124	$\text{Fm}^{3+} + \text{e}^- = \text{Fm}^{2+}$	-1,1
125	$\text{Nb}^{3+} + 3 \text{e}^- = \text{Nb}$	-1,099
126	$\text{Yb}^{3+} + \text{e}^- = \text{Yb}^{2+}$	-1,05
127	$\text{PO}_4^{3-} + 2 \text{H}_2\text{O} + 3 \text{e}^- = \text{HPO}_3^{2-} + 3 \text{OH}^-$	-1,05
128	$\text{In}_2\text{O}_3 + 3 \text{H}_2\text{O} + 6 \text{e}^- = 2 \text{In} + 6 \text{OH}^-$	-1,034
129	$\text{In}(\text{OH})_4^- + 3 \text{e}^- = \text{In} + 4 \text{OH}^-$	-1,007
130	$\text{In}(\text{OH})_3 + 3 \text{e}^- = \text{In} + 3 \text{OH}^-$	-0,99
131	$\text{NpO}_2 + \text{H}_2\text{O} + \text{H}^+ + \text{e}^- = \text{Np}(\text{OH})_3$	-0,962
132	$\text{SnO}_2 + 2 \text{H}_2\text{O} + 4 \text{e}^- = \text{Sn} + 4 \text{OH}^-$	-0,945
133	$\text{Sn}(\text{OH})_6^{2-} + 2 \text{e}^- = \text{HSnO}_2^- + 3 \text{OH}^- + \text{H}_2\text{O}$	-0,93
134	$\text{SO}_4^{2-} + \text{H}_2\text{O} + 2 \text{e}^- = \text{SO}_3^{2-} + 2 \text{OH}^-$	-0,93
135	$\text{Se} + 2 \text{e}^- = \text{Se}^{2-}$	-0,924
136	$\text{Cr}^{2+} + 2 \text{e}^- = \text{Cr}$	-0,913
137	$\text{HSnO}_2^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{Sn} + 3 \text{OH}^-$	-0,909
138	$\text{Ti}^{3+} + \text{e}^- = \text{Ti}^{2+}$	-0,9
139	$\text{P} + 3 \text{H}_2\text{O} + 3 \text{e}^- = \text{PH}_3(\text{g}) + 3 \text{OH}^-$	-0,87
140	$\text{H}_3\text{BO}_3 + 3 \text{H}^+ + 3 \text{e}^- = \text{B} + 3 \text{H}_2\text{O}$	-0,8698
141	$2 \text{NO}_3^- + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{N}_2\text{O}_4 + 4 \text{OH}^-$	-0,85
142	$2 \text{H}_2\text{O} + 2 \text{e}^- = \text{H}_2 + 2 \text{OH}^-$	-0,8277
143	$\text{Cd}(\text{OH})_2 + 2 \text{e}^- = \text{Cd}(\text{Hg}) + 2 \text{OH}^-$	-0,809
144	$\text{SiO} + 2 \text{H}^+ + 2 \text{e}^- = \text{Si} + \text{H}_2\text{O}$	-0,8
145	$\text{Bi} + 3 \text{H}^+ + 3 \text{e}^- = \text{BiH}_3$	-0,8
146	$\text{ZnSO}_4 + 7 \text{H}_2\text{O} + 2 \text{e}^- = \text{Zn}(\text{Hg}) + \text{SO}_4^{2-} \text{ (saturated Z)}$	-0,7993
147	$\text{Te} + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{Te}$	-0,793
148	$\text{CdO} + \text{H}_2\text{O} + 2 \text{e}^- = \text{Cd} + 2 \text{OH}^-$	-0,783
149	$\text{Zn}^{2+} + 2 \text{e}^- = \text{Zn}$	-0,7618
150	$\text{Tl} + \text{e}^- = \text{Tl} + \text{l}^-$	-0,752
151	$\text{Ta}_2\text{O}_5 + 10 \text{H}^+ + 10 \text{e}^- = 2 \text{Ta} + 5 \text{H}_2\text{O}$	-0,75
152	$\text{Cr}^{3+} + 3 \text{e}^- = \text{Cr}$	-0,744
153	$\text{NbO} + 2 \text{H}^+ + 2 \text{e}^- = \text{Nb} + \text{H}_2\text{O}$	-0,733
154	$\text{Zn}^{2+} + 2 \text{e}^- = \text{Zn}(\text{Hg})$	-0,7328

155	$\text{Co(OH)}_2 + 2 \text{e}^- = \text{Co} + 2 \text{OH}^-$	-0,73
156	$\text{Ni(OH)}_2 + 2 \text{e}^- = \text{Ni} + 2 \text{OH}^-$	-0,72
157	$\text{AsO}_4 + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{AsO}_2^- + 4 \text{OH}^-$	-0,71
158	$\text{Ag}_2\text{S} + 2 \text{e}^- = 2 \text{Ag} + \text{S}^{2-}$	-0,691
159	$\text{NbO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{Nb} + 2 \text{H}_2\text{O}$	-0,69
160	$\text{AsO}_2^- + 2 \text{H}_2\text{O} + 3 \text{e}^- = \text{As} + 4 \text{OH}^-$	-0,68
161	$\text{SbO}_2^- + 2 \text{H}_2\text{O} + 3 \text{e}^- = \text{Sb} + 4 \text{OH}^-$	-0,66
162	$\text{TlBr} + \text{e}^- = \text{Tl} + \text{Br}^-$	-0,658
163	$\text{Cd(OH)}_2 + 2 \text{e}^- = \text{Cd} + 2 \text{OH}^-$	-0,658
164	$\text{NbO}_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{NbO} + \text{H}_2\text{O}$	-0,646
165	$\text{Nb}_2\text{O}_5 + 10 \text{H}^+ + 10 \text{e}^- = 2 \text{Nb} + 5 \text{H}_2\text{O}$	-0,644
166	$\text{As} + 3 \text{H}^+ + 3 \text{e}^- = \text{AsH}_3$	-0,608
167	$\text{U}^{4+} + \text{e}^- = \text{U}^{3+}$	-0,607
168	$\text{Ta}^{3+} + 3 \text{e}^- = \text{Ta}$	-0,6
169	$\text{SbO}_3^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{SbO}_2^- + 2 \text{OH}^-$	-0,59
170	$\text{ReO}_4^- + 4 \text{H}_2\text{O} + 7 \text{e}^- = \text{Re} + 8 \text{OH}^-$	-0,584
171	$\text{PbO} + \text{H}_2\text{O} + 2 \text{e}^- = \text{Pb} + 2 \text{OH}^-$	-0,58
172	$2 \text{SO}_3^{2-} + 3 \text{H}_2\text{O} + 4 \text{e}^- = \text{S}_2\text{O}_3^{2-} + 6 \text{OH}^-$	-0,571
173	$\text{TeO}_3^{2-} + 3 \text{H}_2\text{O} + 4 \text{e}^- = \text{Te} + 6 \text{OH}^-$	-0,57
174	$\text{Fe(OH)}_3 + \text{e}^- = \text{Fe(OH)}_2 + \text{OH}^-$	-0,56
175	$\text{TlCl} + \text{e}^- = \text{Tl} + \text{Cl}^-$	-0,5568
176	$\text{Ga}^{3+} + 3 \text{e}^- = \text{Ga}$	-0,549
177	$\text{HPbO}_2^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{Pb} + 3 \text{OH}^-$	-0,537
178	$\text{Sb} + 3 \text{H}^+ + 3 \text{e}^- = \text{SbH}_3$	-0,51
179	$\text{H}_3\text{PO}_2 + \text{H}^+ + 3 \text{e}^- = \text{P} + 2 \text{H}_2\text{O}$	-0,508
180	$\text{TiO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{Ti}^{2+} + 2 \text{H}_2\text{O}$	-0,502
181	$\text{H}_3\text{PO}_3 + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_3\text{PO}_2 + \text{H}_2\text{O}$	-0,499
182	$\text{GaOH}_2^+ + \text{H}^+ + 3 \text{e}^- = \text{Ga} + \text{H}_2\text{O}$	-0,498
183	$\text{ZnOH}^+ + \text{H}^+ + 2 \text{e}^- = \text{Zn} + \text{H}_2\text{O}$	-0,497
184	$\text{NiO}_2 + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{Ni(OH)}_2 + 2 \text{OH}^-$	-0,49
185	$\text{In}^{3+} + \text{e}^- = \text{In}^{2+}$	-0,49
186	$\text{B(OH)}_3 + 7 \text{H}^+ + 8 \text{e}^- = \text{BH}_4^- + 3 \text{H}_2\text{O}$	-0,481
187	$\text{S} + \text{H}_2\text{O} + 2 \text{e}^- = \text{SH}^- + \text{OH}^-$	-0,478
188	$\text{S} + 2 \text{e}^- = \text{S}^{2-}$	-0,47627
189	$\text{PbHPO}_4 + 2 \text{e}^- = \text{Pb} + \text{HPO}_4^{2-}$	-0,465
190	$\text{NO}_2^- + \text{H}_2\text{O} + 3 \text{e}^- = \text{NO} + 2 \text{OH}^-$	-0,46
191	$\text{Bi}_2\text{O}_3 + 3 \text{H}_2\text{O} + 6 \text{e}^- = 2 \text{Bi} + 6 \text{OH}^-$	-0,46
192	$\text{H}_3\text{PO}_3 + 3 \text{H}^+ + 3 \text{e}^- = \text{P} + 3 \text{H}_2\text{O}$	-0,454
193	$\text{Fe}^{2+} + 2 \text{e}^- = \text{Fe}$	-0,447
194	$\text{In}^{3+} + 2 \text{e}^- = \text{In}^+$	-0,443
195	$\text{Ti}_2\text{SO}_4 + 2 \text{e}^- = \text{Ti} + \text{SO}_4^{2-}$	-0,436
196	$2 \text{S} + 2 \text{e}^- = \text{S}_2^{2-}$	-0,42836
197	$\text{Cr}^{3+} + \text{e}^- = \text{Cr}^{2+}$	-0,407
198	$\text{Cd}^{2+} + 2 \text{e}^- = \text{Cd}$	-0,403
199	$\text{In}^{2+} + \text{e}^- = \text{In}^+$	-0,4
200	$\text{Se} + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{Se(aq)}$	-0,399
201	$\text{SeO}_3^{2-} + 3 \text{H}_2\text{O} + 4 \text{e}^- = \text{Se} + 6 \text{OH}^-$	-0,366
202	$\text{PbI}_2 + 2 \text{e}^- = \text{Pb} + 2 \text{I}^-$	-0,365
203	$\text{Eu}^{3+} + \text{e}^- = \text{Eu}^{2+}$	-0,36
204	$\text{Cu}_2\text{O} + \text{H}_2\text{O} + 2 \text{e}^- = 2 \text{Cu} + 2 \text{OH}^-$	-0,36
205	$\text{PbSO}_4 + 2 \text{e}^- = \text{Pb} + \text{SO}_4^{2-}$	-0,3588
206	$\text{Cd}^{2+} + 2 \text{e}^- = \text{Cd(Hg)}$	-0,3521

207	$\text{PbSO}_4 + 2 \text{e}^- = \text{Pb(Hg)} + \text{SO}_4^{2-}$	-0,3505
208	$\text{PbF}_2 + 2 \text{e}^- = \text{Pb} + 2 \text{F}^-$	-0,3444
209	$\text{TlOH} + \text{e}^- = \text{Tl} + \text{OH}^-$	-0,34
210	$\text{In}^{3+} + 3 \text{e}^- = \text{In}$	-0,3382
211	$\text{Tl}^+ + \text{e}^- = \text{Tl}$	-0,336
212	$\text{Tl}^+ + \text{e}^- = \text{Tl(Hg)}$	-0,3338
213	$\text{PbBr}_2 + 2 \text{e}^- = \text{Pb} + 2 \text{Br}^-$	-0,284
214	$\text{Co}^{2+} + 2 \text{e}^- = \text{Co}$	-0,28
215	$\text{H}_3\text{PO}_4 + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_3\text{PO}_3 + \text{H}_2\text{O}$	-0,276
216	$\text{PbCl}_2 + 2 \text{e}^- = \text{Pb} + 2 \text{Cl}^-$	-0,2675
217	$\text{Ni}^{2+} + 2 \text{e}^- = \text{Ni}$	-0,257
218	$\text{V}^{3+} + \text{e}^- = \text{V}^{2+}$	-0,255
219	$\text{V(OH)}_4^+ + 4 \text{H}^+ + 5 \text{e}^- = \text{V} + 4 \text{H}_2\text{O}$	-0,254
220	$\text{CdSO}_4 + 2 \text{e}^- = \text{Cd} + \text{SO}_4^{2-}$	-0,246
221	$\text{V}_2\text{O}_5 + 10 \text{H}^+ + 10 \text{e}^- = 2 \text{V} + 5 \text{H}_2\text{O}$	-0,242
222	$\text{Cu(OH)}_2 + 2 \text{e}^- = \text{Cu} + 2 \text{OH}^-$	-0,222
223	$2 \text{SO}_4^{2-} + 4 \text{H}^+ + 2 \text{e}^- = \text{S}_2\text{O}_6^{2-} + \text{H}_2\text{O}$	-0,22
224	$\text{Mo}^{3+} + 3 \text{e}^- = \text{Mo}$	-0,2
225	$\text{Ga}^+ + \text{e}^- = \text{Ga}$	-0,2
226	$\text{CO}_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{HCOOH}$	-0,199
227	$\text{SnO}_2 + 3 \text{H}^+ + 2 \text{e}^- = \text{SnOH}^+ + \text{H}_2\text{O}$	-0,194
228	$\text{H}_2\text{GeO}_3 + 4 \text{H}^+ + 4 \text{e}^- = \text{Ge} + 3 \text{H}_2\text{O}$	-0,182
229	$2 \text{NO}_2^- + 2 \text{H}_2\text{O} + 4 \text{e}^- = \text{N}_2^{2-} + 4 \text{OH}^-$	-0,18
230	$\text{AgI} + \text{e}^- = \text{Ag} + \text{I}^-$	-0,15224
231	$\text{MoO}_2 + 4 \text{H}^+ + \text{e}^- = \text{Mo} + 4 \text{H}_2\text{O}$	-0,152
232	$\text{O}_2 + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{H}_2\text{O}_2 + 2 \text{OH}^-$	-0,146
233	$\text{In}^+ + \text{e}^- = \text{In}$	-0,14
234	$\text{Sn}^{2+} + 2 \text{e}^- = \text{Sn}$	-0,1375
235	$\text{CrO}_4^{2-} + 4 \text{H}_2\text{O} + 3 \text{e}^- = \text{Cr(OH)}_3 + 5 \text{OH}^-$	-0,13
236	$\text{Pb}^{2+} + 2 \text{e}^- = \text{Pb}$	-0,1262
237	$\text{Pb}^{2+} + 2 \text{e}^- = \text{Pb(Hg)}$	-0,1205
238	$\text{WO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{W} + 2 \text{H}_2\text{O}$	-0,119
239	$\text{GeO}_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{GeO} + \text{H}_2\text{O}$	-0,118
240	$\text{SnO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{Sn} + 2 \text{H}_2\text{O}$	-0,117
241	$\text{P(red)} + 3 \text{H}^+ + 3 \text{e}^- = \text{PH}_3(\text{g})$	-0,111
242	$\text{Md}^{3+} + \text{e}^- = \text{Md}^{2+}$	-0,1
243	$\text{SnO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{Sn}^{2+} + 2 \text{H}_2\text{O}$	-0,094
244	$\text{WO}_3 + 6 \text{H}^+ + 6 \text{e}^- = \text{W} + 3 \text{H}_2\text{O}$	-0,09
245	$\text{Se} + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{Se}$	-0,082
246	$2 \text{Cu(OH)}_2 + 2 \text{e}^- = \text{Cu}_2\text{O} + 2 \text{OH}^- + \text{H}_2\text{O}$	-0,08
247	$\text{O}_2 + \text{H}_2\text{O} + 2 \text{e}^- = \text{HO}_2^- + \text{OH}^-$	-0,076
248	$\text{P(white)} + 3 \text{H}^+ + 3 \text{e}^- = \text{PH}_3(\text{g})$	-0,063
249	$2 \text{H}_2\text{SO}_3 + \text{H}^+ + 2 \text{e}^- = \text{HS}_2\text{O}_4^- + 2 \text{H}_2\text{O}$	-0,056
250	$\text{TiOH}_3^+ + \text{H}^+ + \text{e}^- = \text{Ti}^{3+} + \text{H}_2\text{O}$	-0,055
251	$\text{Ti(OH)}_3 + 2 \text{e}^- = \text{TlOH} + 2 \text{OH}^-$	-0,05
252	$\text{Hg}_2\text{I}_2 + 2 \text{e}^- = 2 \text{Hg} + 2 \text{I}^-$	-0,0405
253	$\text{Fe}^{3+} + 3 \text{e}^- = \text{Fe}$	-0,037
254	$\text{Ag}_2\text{S} + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{Ag} + \text{H}_2\text{S}$	-0,0366
255	$\text{W}_2\text{O}_5 + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{WO}_2 + \text{H}_2\text{O}$	-0,031
256	$2 \text{WO}_3 + 2 \text{H}^+ + 2 \text{e}^- = \text{W}_2\text{O}_5 + \text{H}_2\text{O}$	-0,029
257	$\text{AgCN} + \text{e}^- = \text{Ag} + \text{CN}^-$	-0,017
258	$2 \text{H}^+ + 2 \text{e}^- = \text{H}_2$	0

259	$\text{Ge}^{4+} + 2 \text{e}^- = \text{Ge}^{2+}$	0
260	$\text{CuI}_2 + \text{e}^- = \text{Cu} + 2 \text{I}^-$	0
261	$\text{NO}_3^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{NO}_2^- + 2 \text{OH}^-$	0,01
262	$\text{Ti}_2\text{O}_3 + 3 \text{H}_2\text{O} + 4 \text{e}^- = 2 \text{Ti} + 6 \text{OH}^-$	0,02
263	$\text{WO}_3 + 2 \text{H}^+ + 2 \text{e}^- = \text{WO}_2 + \text{H}_2\text{O}$	0,036
264	$\text{SeO}_4^{2-} + \text{H}_2\text{O} + 2 \text{e}^- = \text{SeO}_3^{2-} + 2 \text{OH}^-$	0,05
265	$\text{UO}_2^{2+} + \text{e}^- = \text{UO}^{2+}$	0,062
266	$\text{Pd}(\text{OH})_2 + 2 \text{e}^- = \text{Pd} + 2 \text{OH}^-$	0,07
267	$\text{AgBr} + \text{e}^- = \text{Ag} + \text{Br}^-$	0,07133
268	$\text{MoO}_3 + 6 \text{H}^+ + 6 \text{e}^- = \text{Mo} + 3 \text{H}_2\text{O}$	0,075
269	$\text{S}_4\text{O}_6^{2-} + 2 \text{e}^- = 2\text{S}_2\text{O}_3^{2-}$	0,08
270	$\text{H}_3\text{Mo}_7\text{O}_{24}^{3-} + 45 \text{H}^+ + 42 \text{e}^- = 7 \text{Mo} + 24 \text{H}_2\text{O}$	0,082
271	$\text{AgSCN} + \text{e}^- = \text{Ag} + \text{SCN}^-$	0,08951
272	$\text{N}_2 + 2 \text{H}_2\text{O} + 6 \text{H}^+ + 6 \text{e}^- = 2 \text{NH}_4\text{OH}$	0,092
273	$\text{HgO} + \text{H}_2\text{O} + 2 \text{e}^- = \text{Hg} + 2 \text{OH}^-$	0,0977
274	$\text{Ir}_2\text{O}_3 + 3 \text{H}_2\text{O} + 6 \text{e}^- = 2 \text{Ir} + 6 \text{OH}^-$	0,098
275	$\text{W}_3 + 3 \text{e}^- = \text{W}$	0,1
276	$[\text{Ru}(\text{NH}_3)_6]^{3+} + \text{e}^- = [\text{Ru}(\text{NH}_3)_6]^{2+}$	0,1
277	$[\text{Co}(\text{NH}_3)_6]^{3+} + \text{e}^- = [\text{Co}(\text{NH}_3)_6]^{2+}$	0,108
278	$\text{Hg}_2\text{O} + \text{H}_2\text{O} + 2 \text{e}^- = 2 \text{Hg} + 2 \text{OH}^-$	0,123
279	$\text{Ge}^{4+} + 4 \text{e}^- = \text{Ge}$	0,124
280	$\text{Hg}_2\text{Br}_2 + 2 \text{e}^- = 2 \text{Hg} + 2 \text{Br}^-$	0,13923
281	$[\text{V}(\text{phen})_3]^{3+} + \text{e}^- = [\text{V}(\text{phen})_3]^{2+}$	0,14
282	$\text{Pt}(\text{OH})_2 + 2 \text{e}^- = \text{Pt} + 2 \text{OH}^-$	0,14
283	$\text{Sn}(\text{OH})_3 + 3 \text{H}^+ + 2 \text{e}^- = \text{Sn}^{2+} + 3 \text{H}_2\text{O}$	0,142
284	$\text{S} + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{S}(\text{aq})$	0,142
285	$\text{Np}^{4+} + \text{e}^- = \text{Np}^{3+}$	0,147
286	$\text{Ag}_4[\text{Fe}(\text{CN})_6] + 4 \text{e}^- = 4 \text{Ag} + [\text{Fe}(\text{CN})_6]^{4-}$	0,1478
287	$2\text{NO}_2^- + 3 \text{H}_2\text{O} + 4 \text{e}^- = \text{N}_2\text{O} + 6 \text{OH}^-$	0,15
288	$\text{Mn}(\text{OH})_3 + \text{e}^- = \text{Mn}(\text{OH})_2 + \text{OH}^-$	0,15
289	$\text{IO}_3^- + 2 \text{H}_2\text{O} + 4 \text{e}^- = \text{IO}^- + 4 \text{OH}^-$	0,15
290	$\text{Sn}^{4+} + 2 \text{e}^- = \text{Sn}^{2+}$	0,151
291	$\text{Sb}_2\text{O}_3 + 6 \text{H}^+ + 6 \text{e}^- = 2 \text{Sb} + 3 \text{H}_2\text{O}$	0,152
292	$\text{Cu}_2 + \text{e}^- = \text{Cu}^+$	0,153
293	$\text{BiOCl} + 2 \text{H}^+ + 3 \text{e}^- = \text{Bi} + \text{Cl}^- + \text{H}_2\text{O}$	0,1583
294	$\text{Fe}_2\text{O}_3 + 4 \text{H}^+ + 2 \text{e}^- = 2\text{FeOH}^+ + \text{H}_2\text{O}$	0,16
295	$\text{BiCl}_4 + 3 \text{e}^- = \text{Bi} + 4 \text{Cl}^-$	0,16
296	$\text{Co}(\text{OH})_3 + \text{e}^- = \text{Co}(\text{OH})_2 + \text{OH}^-$	0,17
297	$\text{SO}_4^{2-} + 4 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{SO}_3 + \text{H}_2\text{O}$	0,172
298	$\text{Bi}^{3+} + 2 \text{e}^- = \text{Bi}^+ \quad 0.2$	
299	$[\text{Ru}(\text{en})_3]^{3+} + \text{e}^- = \text{Ru}(\text{en})_3^{2+}$	0,21
300	$\text{SbO}^+ + 2 \text{H}^+ + 3 \text{e}^- = \text{Sb} + 2 \text{H}_2\text{O}$	0,212
301	$\text{AgCl} + \text{e}^- = \text{Ag} + \text{Cl}^-$	0,22233
302	$[\text{Ru}(\text{H}_2\text{O})_6]^{3+} + \text{e}^- = [\text{Ru}(\text{H}_2\text{O})_6]^{2+}$	0,23
303	$\text{As}_2\text{O}_3 + 6 \text{H}^+ + 6 \text{e}^- = 2 \text{As} + 3 \text{H}_2\text{O}$	0,234
304	$\text{Ge}^{2+} + 2 \text{e}^- = \text{Ge}$	0,24
305	$\text{PbO}_2 + \text{H}_2\text{O} + 2 \text{e}^- = \text{PbO} + 2 \text{OH}^-$	0,247
306	$\text{HAsO}_2 + 3 \text{H}^+ + 3 \text{e}^- = \text{As} + 2 \text{H}_2\text{O}$	0,248
307	$\text{Ru}^{3+} + \text{e}^- = \text{Ru}^{2+}$	0,2487
308	$\text{ReO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{Re} + 2 \text{H}_2\text{O}$	0,2513
309	$\text{IO}_3^- + 3 \text{H}_2\text{O} + 6 \text{e}^- = \text{IO}^- + 6 \text{OH}^-$	0,26
310	$\text{Hg}_2\text{Cl}_2 + 2 \text{e}^- = 2 \text{Hg} + 2 \text{Cl}^-$	0,26808

311	$\text{Tc}^{3+} + \text{e}^- = \text{Tc}^{2+}$	0,3
312	$\text{Re}^{3+} + 3 \text{e}^- = \text{Re}$	0,3
313	$\text{At}_2 + 2 \text{e}^- = 2 \text{At}^-$	0,3
314	$\text{Bi}^{3+} + 3 \text{e}^- = \text{Bi}$	0,308
315	$\text{BiO}^{++} + 2 \text{H}^+ + 3 \text{e}^- = \text{Bi} + \text{H}_2\text{O}$	0,32
316	$\text{UO}_2^{2+} + 4 \text{H}^+ + 2 \text{e}^- = \text{U}^{4+} + 2 \text{H}_2\text{O}$	0,327
317	$\text{ClO}_3^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{ClO}_2^- + 2 \text{OH}^-$	0,33
318	$2\text{HCNO} + 2 \text{H}^+ + 2 \text{e}^- = (\text{CN})_2 + 2 \text{H}_2\text{O}$	0,33
319	$\text{VO}_2^+ + 2 \text{H}^+ + \text{e}^- = \text{V}^{3+} + \text{H}_2\text{O}$	0,337
320	$\text{Cu}^{2+} + 2 \text{e}^- = \text{Cu}$	0,3419
321	$\text{Ag}_2\text{O} + \text{H}_2\text{O} + 2 \text{e}^- = 2 \text{Ag} + 2 \text{OH}^-$	0,342
322	$\text{Cu}^{2+} + 2 \text{e}^- = \text{Cu}(\text{Hg})$	0,345
323	$\text{AgIO}_3 + \text{e}^- = \text{Ag} + \text{IO}_3^-$	0,354
324	$[\text{Fe}(\text{CN})_6]^{3-} + \text{e}^- = [\text{Fe}(\text{CN})_6]^{4-}$	0,358
325	$\text{ClO}_4^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{ClO}_3^- + 2 \text{OH}^-$	0,36
326	$\text{Ag}_2\text{SeO}_3 + 2 \text{e}^- = 2 \text{Ag} + \text{SeO}_4^{2-}$	0,3629
327	$\text{ReO}_4^- + 8 \text{H}^+ + 7 \text{e}^- = \text{Re} + 4 \text{H}_2\text{O}$	0,368
328	$(\text{CN})_2 + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{HCN}$	0,373
329	$\text{Tc}^{2+} + 2 \text{e}^- = \text{Tc}$	0,4
330	$[\text{Ferricinium}]^+ + \text{e}^- = \text{ferrocene}$	0,4
331	$\text{O}_2 + 2 \text{H}_2\text{O} + 4 \text{e}^- = 4 \text{OH}^-$	0,401
332	$\text{AgOCN} + \text{e}^- = \text{Ag} + \text{OCN}^-$	0,41
333	$[\text{RhCl}_6]^{3-} + 3 \text{e}^- = \text{Rh} + 6 \text{Cl}^-$	0,431
334	$\text{Ag}_2\text{CrO}_4 + 2 \text{e}^- = 2 \text{Ag} + \text{CrO}_4^{2-}$	0,447
335	$\text{H}_2\text{SO}_3 + 4 \text{H}^+ + 4 \text{e}^- = \text{S} + 3 \text{H}_2\text{O}$	0,449
336	$\text{Ru}^{2+} + 2 \text{e}^- = \text{Ru}$	0,455
337	$\text{Ag}_2\text{MoO}_4 + 2 \text{e}^- = 2 \text{Ag} + \text{MoO}_4^{2-}$	0,4573
338	$\text{Ag}_2\text{C}_2\text{O}_4 + 2 \text{e}^- = 2 \text{Ag} + \text{C}_2\text{O}_4^{2-}$	0,4647
339	$\text{Ag}_2\text{WO}_4 + 2 \text{e}^- = 2 \text{Ag} + \text{WO}_4^{2-}$	0,466
340	$\text{Ag}_2\text{CO}_3 + 2 \text{e}^- = 2 \text{Ag} + \text{CO}_3^{2-}$	0,47
341	$\text{TeO}_4^- + 8 \text{H}^+ + 7 \text{e}^- = \text{Te} + 4 \text{H}_2\text{O}$	0,472
342	$\text{TcO}_4^- + 8 \text{H}^+ + 7 \text{e}^- = \text{Tc} + 4 \text{H}_2\text{O}$	0,472
343	$\text{IO}^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{I}^- + 2 \text{OH}^-$	0,485
344	$\text{Bi}^+ + \text{e}^- = \text{Bi}$	0,5
345	$\text{ReO}_4^- + 4 \text{H}^+ + 3 \text{e}^- = \text{ReO}_2 + 2 \text{H}_2\text{O}$	0,51
346	$\text{Hg}_2(\text{Ac})_2 + 2 \text{e}^- = 2 \text{Hg} + 2 \text{Ac}^-$	0,51163
347	$\text{Cu}^+ + \text{e}^- = \text{Cu}$	0,521
348	$\text{I}_2 + 2 \text{e}^- = 2\text{I}^-$	0,5355
349	$\text{I}_2^- + 2 \text{e}^- = 3\text{I}^-$	0,536
350	$\text{AgBrO}_3 + \text{e}^- = \text{Ag} + \text{BrO}_3^-$	0,546
351	$\text{MnO}_4^- + \text{e}^- = \text{MnO}_4^{2-}$	0,558
352	$\text{H}_3\text{AsO}_4 + 2 \text{H}^+ + 2 \text{e}^- = \text{HAsO}_2 + 2 \text{H}_2\text{O}$	0,56
353	$\text{S}_2\text{O}_6^{2-} + 4 \text{H}^+ + 2 \text{e}^- = 2\text{H}_2\text{SO}_3$	0,564
354	$\text{AgNO}_2 + \text{e}^- = \text{Ag} + 2 \text{NO}_2^-$	0,564
355	$\text{Te}^{4+} + 4 \text{e}^- = \text{Te}$	0,568
356	$\text{Sb}_2\text{O}_5 + 6 \text{H}^+ + 4 \text{e}^- = 2 \text{SbO}^+ + 3 \text{H}_2\text{O}$	0,581
357	$\text{RuO}_4^- + \text{e}^- = \text{RuO}_4^{2-}$	0,59
358	$[\text{PdCl}_4]^{2-} + 2 \text{e}^- = \text{Pd} + 4 \text{Cl}^-$	0,591
359	$\text{TeO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{Te} + 2 \text{H}_2\text{O}$	0,593
360	$\text{MnO}_4^- + 2 \text{H}_2\text{O} + 3 \text{e}^- = \text{MnO}_2 + 4 \text{OH}^-$	0,595
361	$\text{Rh}_2^{2+} + 2 \text{e}^- = \text{Rh}$	0,6
362	$\text{Rh}^+ + \text{e}^- = \text{Rh}$	0,6

363	$\text{MnO}_4^{2-} + 2 \text{H}_2\text{O} + 2 \text{e}^- = \text{MnO}_2 + 4 \text{OH}^-$	0,6
364	$2 \text{AgO} + \text{H}_2\text{O} + 2 \text{e}^- = \text{Ag}_2\text{O} + 2 \text{OH}^-$	0,607
365	$\text{BrO}_3^- + 3 \text{H}_2\text{O} + 6 \text{e}^- = \text{Br}^- + 6 \text{OH}^-$	0,61
366	$\text{UO}_2^{2+} + 4 \text{H}^+ + \text{e}^- = \text{U}^{4+} + 2 \text{H}_2\text{O}$	0,612
367	$\text{Hg}_2\text{SO}_4 + 2 \text{e}^- = 2 \text{Hg} + \text{SO}_4^{2-}$	0,6125
368	$\text{ClO}_3^- + 3 \text{H}_2\text{O} + 6 \text{e}^- = \text{Cl}^- + 6 \text{OH}^-$	0,62
369	$\text{Hg}_2\text{HPO}_4 + 2 \text{e}^- = 2 \text{Hg} + \text{HPO}_4^{2-}$	0,6359
370	$\text{Ag}(\text{Ac}) + \text{e}^- = \text{Ag} + \text{Ac}^-$	0,643
371	$\text{Sb}_2\text{O}_5 \text{ (valentinite)} + 4 \text{H}^+ + 4 \text{e}^- = \text{Sb}_2\text{O}_3 + 2 \text{H}_2\text{O}$	0,649
372	$\text{Ag}_2\text{SO}_4 + 2 \text{e}^- = 2 \text{Ag} + \text{SO}_4^{2-}$	0,654
373	$\text{ClO}_2^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{ClO}^- + 2 \text{OH}^-$	0,66
374	$\text{Sb}_2\text{O}_5 \text{ (senarmontite)} + 4 \text{H}^+ + 4 \text{e}^- = \text{Sb}_2\text{O}_3 + 2 \text{H}_2\text{O}$	0,671
375	$[\text{PtCl}_6]^{2-} + 2 \text{e}^- = [\text{PtCl}_4]^{2-} + 2 \text{Cl}^-$	0,68
376	$\text{O}_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{O}_2$	0,695
377	$p\text{-benzoquinone} + 2 \text{H}^+ + 2 \text{e}^- = \text{hydroquinone}$	0,6992
378	$\text{H}_3\text{IO}_6 + 2 \text{e}^- = \text{IO}_3^- + 3 \text{OH}^-$	0,7
379	$\text{Ag}_2\text{O}_3 + \text{H}_2\text{O} + 2 \text{e}^- = 2 \text{AgO} + 2 \text{OH}^-$	0,739
380	$\text{H}_2\text{SeO}_3 + 4 \text{H}^+ + 4 \text{e}^- = \text{Se} + 3 \text{H}_2\text{O}$	0,74
381	$\text{Ti}^{3+} + 3 \text{e}^- = \text{Ti}$	0,741
382	$[\text{PtCl}_4]^{2-} + 2 \text{e}^- = \text{Pt} + 4 \text{Cl}^-$	0,755
383	$\text{Rh}^{3+} + 3 \text{e}^- = \text{Rh}$	0,758
384	$\text{Po}^{4+} + 4 \text{e}^- = \text{Po}$	0,76
385	$2 \text{NO} + \text{H}_2\text{O} + 2 \text{e}^- = \text{N}_2\text{O} + 2 \text{OH}^-$	0,76
386	$\text{ClO}_2^- + 2 \text{H}_2\text{O} + 4 \text{e}^- = \text{Cl}^- + 4 \text{OH}^-$	0,76
387	$\text{BrO}^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{Br}^- + 2 \text{OH}^-$	0,761
388	$\text{ReO}_4^- + 2 \text{H}^+ + \text{e}^- = \text{ReO}_3 + \text{H}_2\text{O}$	0,768
389	$[\text{IrCl}_6]^{3-} + 3 \text{e}^- = \text{Ir} + 6 \text{Cl}^-$	0,77
390	$(\text{CNS})_2 + 2 \text{e}^- = 2 \text{CNS}^-$	0,77
391	$\text{Fe}^{3+} + \text{e}^- = \text{Fe}^{2+}$	0,771
392	$\text{AgF} + \text{e}^- = \text{Ag} + \text{F}^-$	0,779
393	$[\text{Fe}(\text{bipy})_2]^{3+} + \text{e}^- = [\text{Fe}(\text{bipy})_2]^{2+}$	0,78
394	$\text{TcO}_4^- + 4 \text{H}^+ + 3 \text{e}^- = \text{TcO}_2 + 2 \text{H}_2\text{O}$	0,782
395	$\text{Hg}_2^{2+} + 2 \text{e}^- = 2 \text{Hg}$	0,7973
396	$\text{Ag}^+ + \text{e}^- = \text{Ag}$	0,7996
397	$[\text{Os}(\text{bipy})_3]^{3+} + \text{e}^- = [\text{Os}(\text{bipy})_3]^{2+}$	0,8
398	$2 \text{NO}_3^- + 4 \text{H}^+ + 2 \text{e}^- = \text{N}_2\text{O}_4 + 2 \text{H}_2\text{O}$	0,803
399	$[\text{Os}(\text{bipy})_2]^{3+} + \text{e}^- = [\text{Os}(\text{bipy})_2]^{2+}$	0,81
400	$\text{ClO}^- + \text{H}_2\text{O} + 2 \text{e}^- = \text{Cl}^- + 2 \text{OH}^-$	0,81
401	$\text{RhOH}_2^+ + \text{H}^+ + 3 \text{e}^- = \text{Rh} + \text{H}_2\text{O}$	0,83
402	$\text{OsO}_4 + 8 \text{H}^+ + 8 \text{e}^- = \text{Os} + 4 \text{H}_2\text{O}$	0,838
403	$\text{Hg}_2^{2+} + 2 \text{e}^- = \text{Hg}$	0,851
404	$\text{AuBr}_4^- + 3 \text{e}^- = \text{Au} + 4 \text{Br}^-$	0,854
405	$\text{SiO}_2 \text{ (quartz)} + 4 \text{H}^+ + 4 \text{e}^- = \text{Si} + 2 \text{H}_2\text{O}$	0,857
406	$[\text{Ru}(\text{CN})_6]^{3-} + \text{e}^- = [\text{Ru}(\text{CN})_6]^{4-}$	0,86
407	$2 \text{HNO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{H}_2\text{N}_2\text{O}_2 + 2 \text{H}_2\text{O}$	0,86
408	$[\text{IrCl}_6]^{2-} + \text{e}^- = [\text{IrCl}_6]^{3-}$	0,8665
409	$\text{N}_2\text{O}_4 + 2 \text{e}^- = 2 \text{NO}_2^-$	0,867
410	$\text{HO}_2^- + \text{H}_2\text{O} + 2 \text{e}^- = 3 \text{OH}^-$	0,878
411	$\text{Po}^{4+} + 2 \text{e}^- = \text{Po}^{2+}$	0,9
412	$2 \text{Hg}_2^{2+} + 2 \text{e}^- = \text{Hg}_2^{2+}$	0,92
413	$\text{NO}_3^- + 3 \text{H}^+ + 2 \text{e}^- = \text{HNO}_2 + \text{H}_2\text{O}$	0,934
414	$\text{Pd}^{2+} + 2 \text{e}^- = \text{Pd}$	0,951



415	$\text{ClO}_2(\text{aq}) + \text{e}^- = \text{ClO}_2^-$	0,954
416	$\text{V}_2\text{O}_5 + 6 \text{H}^+ + 2 \text{e}^- = 2 \text{VO}_2^+ + 3 \text{H}_2\text{O}$	0,957
417	$\text{NO}_3^- + 4 \text{H}^+ + 3 \text{e}^- = \text{NO} + 2 \text{H}_2\text{O}$	0,957
418	$\text{AuBr}_2^- + \text{e}^- = \text{Au} + 2 \text{Br}^-$	0,959
419	$\text{HNO}_2 + \text{H}^+ + \text{e}^- = \text{NO} + \text{H}_2\text{O}$	0,983
420	$\text{HIO} + \text{H}^+ + 2 \text{e}^- = \text{I}^- + \text{H}_2\text{O}$	0,987
421	$\text{VO}_2^+ + 2 \text{H}^+ + \text{e}^- = \text{VO}^{2+} + \text{H}_2\text{O}$	0,991
422	$\text{V}(\text{OH})_4^+ + 2 \text{H}^+ + \text{e}^- = \text{VO}_2^+ + 3 \text{H}_2\text{O}$	1
423	$\text{RuO}_4 + \text{e}^- = \text{RuO}_4^-$	1
424	$\text{PtO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{Pt} + 2 \text{H}_2\text{O}$	1
425	$\text{AuCl}_4^- + 3 \text{e}^- = \text{Au} + 4 \text{Cl}^-$	1,002
426	$\text{Pu}^{4+} + \text{e}^- = \text{Pu}^{3+}$	1,006
427	$\text{PtO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{PtO} + 2 \text{H}_2\text{O}$	1,01
428	$\text{H}_6\text{TeO}_6 + 2 \text{H}^+ + 2 \text{e}^- = \text{TeO}_2 + 4 \text{H}_2\text{O}$	1,02
429	$\text{OsO}_4 + 4 \text{H}^+ + 4 \text{e}^- = \text{OsO}_2 + 2 \text{H}_2\text{O}$	1,02
430	$[\text{Fe}(\text{bipy})_3]^{3+} + \text{e}^- = [\text{Fe}(\text{bipy})_3]^{2+}$	1,03
431	$\text{Hg}(\text{OH})_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{Hg} + 2 \text{H}_2\text{O}$	1,034
432	$\text{N}_2\text{O}_4 + 4 \text{H}^+ + 4 \text{e}^- = 2 \text{NO} + 2 \text{H}_2\text{O}$	1,035
433	$\text{RuO}_4 + 8 \text{H}^+ + 8 \text{e}^- = \text{Ru} + 4 \text{H}_2\text{O}$	1,038
434	$[\text{Fe}(\text{phen})_3]^{3+} + \text{e}^- = [\text{Fe}(\text{phen})_3]^{2+} \text{ (1 molar H}_2\text{SO}_4)$	1,06
435	$\text{PuO}_2(\text{OH})_2 + \text{H}^+ + \text{e}^- = \text{PuO}_2\text{OH} + \text{H}_2\text{O}$	1,062
436	$\text{N}_2\text{O}_4 + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{NH}_2\text{O}$	1,065
437	$\text{Br}_2(\text{liq}) + 2 \text{e}^- = 2 \text{Br}^-$	1,066
438	$\text{IO}_3^- + 6 \text{H}^+ + 6 \text{e}^- = \text{I}^- + 3 \text{H}_2\text{O}$	1,085
439	$\text{Br}_2(\text{aq}) + 2 \text{e}^- = 2 \text{Br}^-$	1,0873
440	$\text{Pu}^{5+} + \text{e}^- = \text{Pu}^{4+}$	1,099
441	$\text{Cu}^{2+} + 2 \text{CN}^- + \text{e}^- = [\text{Cu}(\text{CN})_2]^-$	1,103
442	$\text{RuO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{Ru}^{2+} + 2 \text{H}_2\text{O}$	1,12
443	$[\text{Fe}(\text{phen})_3]^{3+} + \text{e}^- = [\text{Fe}(\text{phen})_3]^{2+}$	1,147
444	$\text{SeO}_4^{2-} + 4 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{SeO}_3 + \text{H}_2\text{O}$	1,151
445	$\text{ClO}_3^- + 2 \text{H}^+ + \text{e}^- = \text{ClO}_2 + \text{H}_2\text{O}$	1,152
446	$\text{Ir}^{3+} + 3 \text{e}^- = \text{Ir}$	1,156
447	$\text{Pt}^{2+} + 2 \text{e}^- = \text{Pt}$	1,18
448	$\text{ClO}_4^- + 2 \text{H}^+ + 2 \text{e}^- = \text{ClO}_3^- + \text{H}_2\text{O}$	1,189
449	$2 \text{IO}_3^- + 12 \text{H}^+ + 10 \text{e}^- = \text{I}_2 + 6 \text{H}_2\text{O}$	1,195
450	$\text{PTOH}^+ + \text{H}^+ + 2 \text{e}^- = \text{Pt} + \text{H}_2\text{O}$	1,2
451	$\text{ClO}_3^- + 3 \text{H}^+ + 2 \text{e}^- = \text{HClO}_2 + \text{H}_2\text{O}$	1,214
452	$\text{MnO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{Mn}^{2+} + 2 \text{H}_2\text{O}$	1,224
453	$\text{O}_2 + 4 \text{H}^+ + 4 \text{e}^- = 2 \text{H}_2\text{O}$	1,229
454	$\text{Cr}_2\text{O}_7^{2-} + 14 \text{H}^+ + 6 \text{e}^- = 2 \text{Cr}^{3+} + 7 \text{H}_2\text{O}$	1,232
455	$[\text{Ru}(\text{bipy})_3]^{3+} + \text{e}^- = [\text{Ru}(\text{bipy})_3]^{2+}$	1,24
456	$\text{O}_3 + \text{H}_2\text{O} + 2 \text{e}^- = \text{O}_2 + 2 \text{OH}^-$	1,24
457	$\text{Tl}^{3+} + 2 \text{e}^- = \text{Tl}^+$	1,252
458	$\text{N}_5^+ + 3 \text{H}^+ + 2 \text{e}^- = 2 \text{NH}_4^+$	1,275
459	$\text{ClO}_2 + \text{H}^+ + \text{e}^- = \text{HClO}_2$	1,277
460	$[\text{PdCl}_6]^{2-} + 2 \text{e}^- = [\text{PdCl}_4]^{2-} + 2 \text{Cl}^-$	1,288
461	$2 \text{HNO}_2 + 4 \text{H}^+ + 4 \text{e}^- = \text{N}_2\text{O} + 3 \text{H}_2\text{O}$	1,297
462	$\text{AuOH}_2^+ + \text{H}^+ + 2 \text{e}^- = \text{Au}^+ + \text{H}_2\text{O}$	1,32
463	$\text{PuO}_2(\text{OH})_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{Pu}(\text{OH})_4$	1,325
464	$\text{HBrO} + \text{H}^+ + 2 \text{e}^- = \text{Br}^- + \text{H}_2\text{O}$	1,331
465	$\text{HCrO}_4^- + 7 \text{H}^+ + 3 \text{e}^- = \text{Cr}^{3+} + 4 \text{H}_2\text{O}$	1,35
466	$\text{Cl}_2(\text{g}) + 2 \text{e}^- = 2 \text{Cl}^-$	1,35827

467	$\text{ClO}_4^- + 8 \text{H}^+ + 8 \text{e}^- = \text{Cl}^- + 4 \text{H}_2\text{O}$	1,389
468	$\text{ClO}_4^- + 8 \text{H}^+ + 7 \text{e}^- = 1/2 \text{Cl}_2 + 4 \text{H}_2\text{O}$	1,39
469	$\text{RuO}_4 + 6 \text{H}^+ + 4 \text{e}^- = \text{Ru}(\text{OH})_2^{2+} + 2 \text{H}_2\text{O}$	1,4
470	$\text{No}_3^+ + \text{e}^- = \text{No}_2^+$	1,4
471	$\text{Au}_3^+ + 2 \text{e}^- = \text{Au}^+$	1,401
472	$2\text{NH}_3\text{OH}^+ + \text{H}^+ + 2 \text{e}^- = \text{N}_2\text{H}_5^+ + 2 \text{H}_2\text{O}$	1,42
473	$\text{BrO}_3^- + 6 \text{H}^+ + 6 \text{e}^- = \text{Br}^- + 3 \text{H}_2\text{O}$	1,423
474	$2\text{HIO} + 2 \text{H}^+ + 2 \text{e}^- = \text{I}_2 + 2 \text{H}_2\text{O}$	1,439
475	$\text{Au}(\text{OH})_3 + 3 \text{H}^+ + 3 \text{e}^- = \text{Au} + 3 \text{H}_2\text{O}$	1,45
476	$\text{ClO}_3^- + 6 \text{H}^+ + 6 \text{e}^- = \text{Cl}^- + 3 \text{H}_2\text{O}$	1,451
477	$\text{PbO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{Pb}_2^{2+} + 2 \text{H}_2\text{O}$	1,455
478	$\text{ClO}_3^- + 6 \text{H}^+ + 5 \text{e}^- = 1/2 \text{Cl}_2 + 3 \text{H}_2\text{O}$	1,47
479	$\text{CrO}_2 + 4 \text{H}^+ + \text{e}^- = \text{Cr}_3^+ + 2 \text{H}_2\text{O}$	1,48
480	$\text{HClO} + \text{H}^+ + 2 \text{e}^- = \text{Cl}^- + \text{H}_2\text{O}$	1,482
481	$\text{BrO}_3^- + 6 \text{H}^+ + 5 \text{e}^- = 1/2 \text{Br}_2 + 3 \text{H}_2\text{O}$	1,482
482	$\text{Mn}_2\text{O}_3 + 6 \text{H}^+ + \text{e}^- = \text{Mn}^{2+} + 3 \text{H}_2\text{O}$	1,485
483	$\text{HO}_2 + \text{H}^+ + \text{e}^- = \text{H}_2\text{O}_2$	1,495
484	$\text{Au}_3^+ + 3 \text{e}^- = \text{Au}$	1,498
485	$\text{PtO}_3 + 4 \text{H}^+ + 2 \text{e}^- = \text{Pt}(\text{OH})_2^{2+} + \text{H}_2\text{O}$	1,5
486	$\text{MnO}_4^- + 8 \text{H}^+ + 5 \text{e}^- = \text{Mn}^{2+} + 4 \text{H}_2\text{O}$	1,507
487	$\text{Mn}^{3+} + \text{e}^- = \text{Mn}^{2+}$	1,5415
488	$\text{HClO}_2 + 3 \text{H}^+ + 4 \text{e}^- = \text{Cl}^- + 2 \text{H}_2\text{O}$	1,57
489	$\text{HBrO} + \text{H}^+ + \text{e}^- = 1/2 \text{Br}_2(\text{aq}) + \text{H}_2\text{O}$	1,574
490	$2 \text{NO} + 2 \text{H}^+ + 2 \text{e}^- = \text{N}_2\text{O} + \text{H}_2\text{O}$	1,591
491	$\text{Bi}_2\text{O}_4 + 4 \text{H}^+ + 2 \text{e}^- = 2 \text{BiO}^+ + 2 \text{H}_2\text{O}$	1,593
492	$\text{HBrO} + \text{H}^+ + \text{e}^- = 1/2 \text{Br}_2(\text{liq}) + \text{H}_2\text{O}$	1,596
493	$\text{H}_5\text{IO}_6 + \text{H}^+ + 2 \text{e}^- = \text{IO}_3^- + 3 \text{H}_2\text{O}$	1,601
494	$\text{HClO} + \text{H}^+ + \text{e}^- = 1/2 \text{Cl}_2 + \text{H}_2\text{O}$	1,611
495	$\text{HClO}_2 + 3 \text{H}^+ + 3 \text{e}^- = 1/2 \text{Cl}_2 + 2 \text{H}_2\text{O}$	1,628
496	$\text{HClO}_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{HClO} + \text{H}_2\text{O}$	1,645
497	$\text{Bk}_4^+ + \text{e}^- = \text{Bk}_3^+$	1,67
498	$\text{NiO}_2 + 4 \text{H}^+ + 2 \text{e}^- = \text{Ni}^{2+} + 2 \text{H}_2\text{O}$	1,678
499	$\text{MnO}_4^- + 4 \text{H}^+ + 3 \text{e}^- = \text{MnO}_2 + 2 \text{H}_2\text{O}$	1,679
500	$\text{PbO}_2 + \text{SO}_4^{2-} + 4 \text{H}^+ + 2 \text{e}^- = \text{PbSO}_4 + 2 \text{H}_2\text{O}$	1,6913
501	$\text{Au}^+ + \text{e}^- = \text{Au}$	1,692
502	$\text{PtO}_3 + 2 \text{H}^+ + 2 \text{e}^- = \text{PtO}_2 + \text{H}_2\text{O}$	1,7
503	$\text{CeOH}_3^+ + \text{H}^+ + \text{e}^- = \text{Ce}^{3+} + \text{H}_2\text{O}$	1,715
504	$\text{Ce}_4^+ + \text{e}^- = \text{Ce}^{3+}$	1,72
505	$\text{N}_2\text{O} + 2 \text{H}^+ + 2 \text{e}^- = \text{N}_2 + \text{H}_2\text{O}$	1,766
506	$\text{H}_2\text{O}_2 + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{H}_2\text{O}$	1,776
507	$\text{Au}_2^+ + \text{e}^- = \text{Au}^+$	1,8
508	$\text{Ag}_3^+ + \text{e}^- = \text{Ag}_2^+$	1,8
509	$\text{Ag}_2\text{O}_2 + 4 \text{H}^+ + \text{e}^- = 2 \text{Ag} + 2 \text{H}_2\text{O}$	1,802
510	$\text{Ag}_3^+ + 2 \text{e}^- = \text{Ag}^+$	1,9
511	$\text{Co}_3^+ + \text{e}^- = \text{Co}^{2+}$	1,92
512	$\text{Ag}_2^+ + \text{e}^- = \text{Ag}^+$	1,98
513	$\text{Cu}_2\text{O}_3 + 6 \text{H}^+ + \text{e}^- = \text{Cu}_2^{2+} + 3 \text{H}_2\text{O}$	2
514	$\text{S}_2\text{O}_8^{2-} + 2 \text{e}^- = 2\text{SO}_4^{2-}$	2,01
515	$\text{OH}^- + \text{e}^- = \text{OH}^-$	2,02
516	$\text{HFeO}_4^- + 8 \text{H}^+ + 3 \text{e}^- = \text{Fe}^{3+} + 4 \text{H}_2\text{O}$	2,07
517	$\text{O}_3 + 2 \text{H}^+ + 2 \text{e}^- = \text{O}_2 + \text{H}_2\text{O}$	2,076
518	$\text{HFeO}_4^- + 4 \text{H}^+ + 3 \text{e}^- = \text{FeOOH} + 2 \text{H}_2\text{O}$	2,08

519	$\text{HFeO}_4^- + 2 \text{H}^+ + 3 \text{e}^- = \text{Fe}_2\text{O}_3 + \text{H}_2\text{O}$	2,09
520	$\text{XeO}_3 + 6 \text{H}^+ + 6 \text{e}^- = \text{Xe} + 3 \text{H}_2\text{O}$	2,1
521	$\text{S}_2\text{O}_8^{2-} + 2 \text{H}^+ + 2 \text{e}^- = 2\text{HSO}_4^-$	2,123
522	$\text{F}_2 + 2 \text{H}^+ + 4 \text{e}^- = \text{H}_2\text{O} + 2 \text{F}^-$	2,153
523	$\text{FeO}_4^{2-} + 8 \text{H}^+ + 3 \text{e}^- = \text{Fe}^{3+} + 4 \text{H}_2\text{O}$	2,2
524	$\text{Al}(\text{OH})_4^- + 3 \text{e}^- = \text{Al} + 4\text{OH}^-$	2,328
525	$\text{Cu}^{3+} + \text{e}^- = \text{Cu}^{2+}$	2,4
526	$\text{H}_4\text{XeO}_6 + 2 \text{H}^+ + 2 \text{e}^- = \text{XeO}_3 + 3 \text{H}_2\text{O}$	2,42
527	$\text{O}(\text{g}) + 2 \text{H}^+ + 2 \text{e}^- = \text{H}_2\text{O}$	2,421
528	$\text{Am}^{4+} + \text{e}^- = \text{Am}^{3+}$	2,6
529	$\text{H}_2\text{N}_2\text{O}_2 + 2 \text{H}^+ + 2 \text{e}^- = \text{N}_2 + 2 \text{H}_2\text{O}$	2,65
530	$\text{F}_2 + 2 \text{e}^- = 2\text{F}^-$	2,866
531	$\text{Cm}^{4+} + \text{e}^- = \text{Cm}^{3+}$	3
532	$\text{F}_2 + 2 \text{H}^+ + 2 \text{e}^- = 2 \text{HF}$	3,053
533	$\text{Tb}^{4+} + \text{e}^- = \text{Tb}^{3+}$	3,1
534	$\text{Pr}^{4+} + \text{e}^- = \text{Pr}^{3+}$	3,2
535	$\text{Cf}^{4+} + \text{e}^- = \text{Cf}^{3+}$	3,3
536	$\text{XeF} + \text{e}^- = \text{Xe} + \text{F}^-$	3,4